

# Oils and Hydrocarbon Source Rocks

from the Forest City Basin, Northeastern Kansas, Northwestern Missouri, Southwestern Iowa, and Southeastern Nebraska

Joseph R. Hatch and K. David Newell

## Geochemistry of Oils and Hydrocarbon Source Rocks from the Forest City Basin, Northeastern Kansas, Northwestern Missouri, Southwestern Iowa, and Southeastern Nebraska

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#### Abstract

Comparisons of saturated hydrocarbon and terpane (m/z = 191) distributions for 26 Forest City basin oils demonstrate three geochemically distinct groups. Group 1 oils (n = 18) are produced from Middle and Upper Ordovician sandstones and limestones and Silurian-Devonian dolomites in anticlines along the north-northeast-south-southwest-trending basin axis close to the Humboldt fault, and from the Vassar and Easton-McLouth complex of fields on its eastern flank. Group 2 oils (n = 7) are produced from stratigraphic and combination structural-stratigraphic traps in lenticular Middle and Upper Pennsylvanian sandstones in fields extending south-southwestward from the Kansas City region into the Cherokee platform, and from the Yaege field on the Nemaha uplift to the west of the Humboldt fault. The Group 3 oil is produced from the Kansas City Group in the Davis Ranch field along the axis of the basin.

Rock-Eval® and organic-carbon analyses of 123 core, sidewall core, cuttings, and mine samples show good hydrocarbon source-rock potential for Middle Ordovician, Upper Devonian-Lower Mississippian, and Middle and Upper Pennsylvanian intervals. Organicmatter thermal maturity for all intervals generally increases from north to south. Organic matter in lower Paleozoic strata is generally marginally mature to mature, whereas, organic matter in Pennsylvanian strata is immature to marginally mature. Comparison of saturated hydrocarbon and terpane (m/z = 191) distributions from rock extracts with those of the oils show that source rocks for the Group 1 oils are shales in the Middle Ordovician Simpson Group; Group 2 oils, the Upper Devonian-Lower Mississippian Chattanooga Shale, and Group 3 oil, marine black shales within the Middle and Upper Pennsylvanian section.

## Introduction

The Forest City basin is a shallow late Paleozoic interior cratonic basin occupying the northeastern portion of Kansas and parts of adjoining Missouri, Iowa, and Nebraska (fig. 1). It is an asymmetric structural depression with a broad eastern flank. The basin axis is situated very close to the Humboldt fault, which separates the basin from the Nemaha uplift to the west. The northern boundary with the Iowa shelf is defined by the Thurman-Redfield fault zone in southwestern Iowa, whereas, the southern boundary with the Cherokee basin is the low-lying Bourbon arch. The eastern boundary is poorly defined and has been placed arbitrarily along the line marking the outcrop of the Middle Pennsylvanian Cherokee Group (Wells, 1971). Maximum depth to the Precambrian surface is about 5,200 ft (1,585 m) in Page County, Iowa (Newell et al., 1987).

The U.S. Geological Survey (1995) and Gautier et al. (1996) estimated mean amounts of undiscovered, technically recoverable conventional hydrocarbons for the Forest City basin to be about 20 million barrels of crude oil, about 70 billion cubic feet of gas, and less than 10 million barrels of natural gas liquids.

Two oil- and gas-production trends dominate the Forest City basin. On the eastern flank of the basin, a broad belt of oil and gas fields extends south-southwestward from the Kansas City region and continues into the Cherokee basin in southeastern Kansas and northeastern Oklahoma. This trend is characterized by stratigraphic and combination structural-stratigraphic traps developed in lenticular Middle and Upper Pennsylvanian sandstones of

the Cherokee, Marmaton, and Pleasanton Groups (Jewett, 1954; Newell et al., 1987). Early production records are not accurate, but fields in this trend have probably produced in excess of 74 million barrels of oil and 6.6 billion cubic feet of gas (Newell et al., 1987).

The second production trend is defined by a series of fields that occur in anticlines present along the northnortheast-south-southwest-trending axis of the basin (fig. 1). Most production in this western trend comes from lower Paleozoic reservoirs including Middle Ordovician Simpson Group sandstones, Upper Ordovician Viola Limestone, and Silurian-Devonian "Hunton" dolomites (Newell et al., 1987). Most reservoirs in this production trend are in relatively deep structural positions along the axis of the Forest City basin. About 32 million barrels of oil have been recovered from fields in this western trend (Newell et al., 1987).

A third hydrocarbon trend is present in western Missouri and eastern Kansas on the southern and eastern basin margins. This trend is characterized by the occurrence of low-gravity, high-viscosity oil in discontinuous shallow Pennsylvanian sand bodies at depths of less than 200 ft (60 m) (Wells and Anderson, 1968; Wells, 1979; Ebanks and Weber, 1987). In-place oil in these shallow deposits in the Forest City basin and Cherokee platform to the south has been estimated to be about two billion barrels (Wells, 1979). Except for local heavy oil produced from stripper wells or through enhanced-recovery techniques (Ebanks and Weber, 1987), production from these deposits currently is not significant. This third trend will not be discussed further in this report.

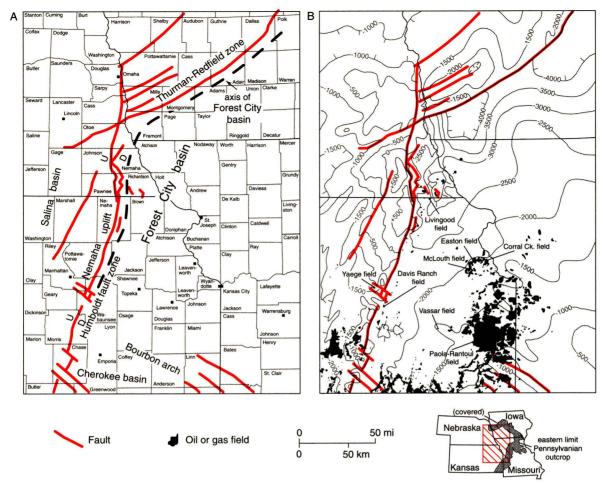


FIGURE 1—Forest City Basin index maps. Basement structure contours (in feet) are from Burchett et al. (1983b); oil and gas fields are from Burchett et al. (1983a) and Paul et al. (1988).

This paper provides a summary of the geochemistry of oils and potential hydrocarbon-source rocks from the Forest City basin and adjacent areas of the midcontinent. This summary includes (1) a brief review of previous studies characterizing the geochemistry of oils and hydrocarbon-source rocks; (2) compilations of new and previously published geochemical data characterizing (a) the

oils from the two identified major production trends in the basin, and (b) potential hydrocarbon-source rocks, including representative samples from formations of Middle and Upper Ordovician, Upper Devonian–Lower Mississippian, and Middle and Upper Pennsylvanian age; and (3) analyses of the available data which identify three different oil groups and correlate these oils with their hydrocarbon-source rocks.

## Previous Geochemical Studies

#### Oils

Previous studies listing geochemical analyses of oils from the Forest City basin and adjacent areas in the midcontinent include Reed et al. (1986), Hatch et al. (1987, 1989), Longman and Palmer (1987), and Newell et al. (1987). These studies clearly document the presence of at least two different oil groups in the Forest City basin. One of these groups, the "Ordovician" oils, is primarily produced from Middle and Upper Ordovician sandstones and limestones, and Silurian–Devonian "Hunton" dolomites in anticlines along the north-northeast–south-southwest-trending basin axis close to the Humboldt fault

(Newell et al., 1987). Oils from this group also are produced from Mississippian limestone and/or Pennsylvanian sandstone reservoirs in at least three fields (Easton and Corral Creek fields, Leavenworth County, Kansas, and Vassar field, Osage County, Kansas) on the eastern side of the Forest City basin. The second oil group is primarily produced from lenticular Middle and Upper Pennsylvanian sandstones in the production trend on the eastern flank of the basin.

Oils similar to those from the first group also are produced from Middle and Upper Ordovician sandstones, limestones, and shales in fields on the southern margin of the Salina basin (Marion, Saline, McPherson, and Harvey counties, Kansas) southwest of the Forest City basin, and from Middle Ordovician dolomite in the Keota Dome field, Washington County, Iowa, northeast of the Forest City basin (Hatch et al., 1987, 1991). Oils similar to those from the second group are produced from lenticular sandstone reservoirs in the Middle Pennsylvanian Cherokee Group in the Cherokee basin in southeastern Kansas and northeastern Oklahoma (Hatch et al., 1989).

Gas-chromatographic characteristics of the Ordovician oils include a relatively high abundance of nalkanes with carbon numbers less than 20, a strong predominance of odd-numbered n-alkanes between  $C_{10}$  and C<sub>20</sub>, and relatively small amounts of branched and cyclic alkanes (Reed et al., 1986; Hatch et al., 1987; Longman and Palmer, 1987). Gas-chromatographic characteristics of the second oil group include regularly decreasing amounts of n-alkanes with increased carbon number, no odd-carbon predominance, and relatively abundant branched and cyclic compounds.

The "Ordovician oils" are characterized by unusually wide ranges in the carbon-isotope compositions of the saturated and aromatic hydrocarbon fractions. Analyses for 10 oils from the Forest City basin listed in Hatch et al. (1987, table 1) and for six oils listed in Longman and Palmer (1987, table 2) show  $\delta^{13}C_{sat}$  ranging from -26.9 to  $-31.3^{\circ}/_{00}$ , and  $\delta^{13}C_{arom}$  from -26.2 to  $-31.5^{\circ}/_{00}$ . Similar wide ranges in oil carbon-isotope compositions were noted for seven "Ordovician" oils collected from the southern margin of the Salina basin by Hatch et al. (1987). There,  $\delta^{13}C_{sat}$  ranges from -24.9 to -30.90/oo, and  $\delta^{13}C_{arom}$  from -24.3 to -31.0% For the "Ordovician" oil from the Keota Dome field in Iowa,  $\delta^{13}C_{sat}$  is -32.1º/oo, and  $\delta^{13}C_{arom}$ is -31.5% (Hatch et al., 1987). In contrast, for seven samples of the second group of oils collected from the Cherokee basin, Hatch et al. (1989) document a much smaller range in carbon-isotope compositions. There,  $\delta^{13}C_{sat}$  ranges from -29.2 to -30.6%, and  $\delta^{13}C_{arom}$  from -28.2 to -29.4º/oo.

Terpane distributions (from GCMS) for the "Ordovician" oils from the Forest City basin are characterized by nearly equal amounts of C<sub>19</sub>, C<sub>20</sub>, C<sub>21</sub>, C<sub>23</sub>, and  $C_{24}$  tricyclic terpanes and relatively large amounts of  $C_{24}$ tetracyclic terpane (Longman and Palmer, 1987). Terpane distributions for oils from the second group from the Cherokee basin are characterized by low relative amounts of C<sub>19</sub> and C<sub>20</sub> tricyclic and C<sub>24</sub> tetracyclic terpanes and high relative amounts of C23 through C28 tricyclic terpanes (Hatch et al., 1989).

## **Organic-matter-rich Rocks**

Previous studies characterizing hydrocarbon sourcerock potential and extract geochemistry of rock intervals from the Forest City basin and adjacent areas of the midcontinent region have focused on organic-matter-rich intervals in the Middle Ordovician Simpson Group and correlative units (Newell et al., 1987; Hatch et al., 1987,

- 1991: Jacobson et al., 1988), the Upper Devonian-Lower Mississippian Chattanooga Shale (Hatch et al., 1989), and on Middle and Upper Pennsylvanian, thin, widespread, marine black shales (Baker, 1962; James, 1970; James and Baker, 1972; Hatch et al., 1984, 1989; Wenger et al., 1988; Ece, 1989; Hatch and Leventhal, 1992, 1997). Newell et al. (1987) modeled organic-matter thermal maturity of the Paleozoic section in the Forest City basin utilizing timetemperature index calculations (after Waples, 1980). Brief summaries of these studies follow:
- (1) Newell et al. (1987, fig. 3C) show the saturated hydrocarbon distribution from an extract of a shale sample (well identification and depth interval not indicated) from the Simpson Group in Wabaunsee County, Kansas. This distribution is characterized by a relatively high abundance of *n*-alkanes with carbon numbers less than 20, a strong predominance of odd-numbered n-alkanes between C<sub>10</sub> and C<sub>20</sub>, and relatively small amounts of branched and cyclic alkanes.
- (2) Hatch et al. (1987, table 3), Jacobson et al. (1988, table 1), and Hatch et al. (1991, table 24-2) list organic carbon contents and Rock-Eval® pyrolysis results for 18 core samples from the Middle Ordovician Glenwood, Platteville, and Decorah formations, Washington County, Iowa. For these samples, total organic carbon content (TOC) ranges from 0.13 to 41.4 weight %; hydrogen index from 69 to 1,000 mg/g, and production index from 0.01 to 0.07 (for samples with >2.0 % TOC). Analyses for a second set of 15 core samples from Middle Ordovician rocks further east in Jackson County, Iowa, show similar results (Hatch et al., 1987, table 3; Jacobson et al., 1988, table 1; Hatch et al., 1991, table 24-2). Organic matter,  $\delta^{13}$ C for the core samples from Washington County, Iowa, systematically shifts from -32.2 to -22.7%, whereas, for the core samples from Jackson County, Iowa, organic matter  $\delta^{13}$ C systematically shifts from -31.9 to -23.5% oo, then back to -29.1% Hatch et al. (1991) conclude that organic carbon and hydrogen indices for these samples demonstrate excellent hydrocarbon source-rock potential for Simpson Group-equivalent rocks in eastern Iowa, whereas, the low production indices (<0.10) indicate immaturity with respect to petroleum generation.
- (3) Hatch et al. (1987, table 2) list  $\delta^{13}$ C of saturated and aromatic hydrocarbon fractions from extracts of three samples of Simpson Group rocks from Wabaunsee County, Kansas, and nine samples from the Glenwood, Platteville, and Decorah formations in Washington and Jackson counties, Iowa.  $\delta^{13}$ C<sub>sat</sub> for these samples ranges from -24.9 to -31.3º/oo;  $\delta^{13}C_{arom}$  from -23.7 to -31.9º/oo.
- (4) Jacobson et al. (1988) demonstrate the presence of two different organic-matter assemblages in cores of Middle Ordovician rocks from Washington and Jackson counties, Iowa. Saturated hydrocarbon distribution for samples with their "assemblage A" organic matter (1,007.8

- ft, E. M. Greene #1 well, their fig. 5C) is very similar to saturated hydrocarbon distributions of the Ordovician oils from the Forest City basin shown by previous researchers.
- (5) Jacobson et al. (1988, table 1) show that for nine core samples from the Middle Ordovician St. Peter Sandstone, and Glenwood, Platteville, and Decorah formations from Plymouth County, Iowa, to the north of the Forest City basin, organic-carbon contents range from 0.4 to 9.3%; hydrogen indices, 43 to 780 mg/g, and production index, <0.01 to 0.03.
- (6) Hatch et al. (1989) list analyses of 13 samples from three partial cores of the Chattanooga Shale in Greenwood and Wabaunsee counties, Kansas. These analyses show that organic-carbon contents range from 1.9 to 5.0%, genetic potential, 5.6 to 22 mg/g, hydrogen indices, 240 to 390 mg/g, and pristane/phytane (n=2) were both 1.6.  $T_{max}$  for these samples ranges from 434° to 445°C, which indicates organic matter in these samples is marginally mature to mature with respect to petroleum generation. Hatch et al. (1989) utilize these data and compare  $\delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  measurements, saturated hydrocarbon distributions, and m/z=191 and m/z=217 fragmentograms from oils and shale extracts to determine that the Chattanooga Shale is the source interval for the oil produced from the Cherokee Group.
- (7) Baker (1962) reports organic-carbon contents, amounts of extractable hydrocarbons, and saturated hydrocarbon/aromatic hydrocarbon ratios of core samples from the Middle Pennsylvanian Cherokee Group on the Cherokee basin in southeastern Kansas and northeastern Oklahoma. He found that the saturated hydrocarbon/ aromatic hydrocarbon ratios from the marine shales and limestones were similar to the oils produced from the Cherokee Group.
- (8) James (1970) lists organic-carbon contents and documents variations in saturated hydrocarbon/aromatic hydrocarbon ratios for samples of the Middle Pennsylvanian Excello shale member of the Fort Scott Limestone (upper part of the Cherokee Group) collected from the eastern midcontinent and Illinois basin regions. James and Baker (1972) proposed that the observed variations in saturated to aromatic hydrocarbon ratios in this shale were controlled by differences in the relative proportions of marine and terrestrial organic matter.
- (9) Hatch et al. (1984) collected 247 samples from the Cherokee Group and Marmaton Group lithologies from cores and coal mines at 21 locations in southeastern Iowa, Missouri, and from the Cherokee platform in southeastern Kansas and northeastern Oklahoma. They list organic-carbon contents and Rock-Eval® results for the 247 samples, vitrinite- reflectance values (R<sub>o</sub>) for 19 coal samples, extractable organic-matter compositions for 77

- samples, and  $\delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  from extracts of 18 coal and shale samples.
- (10) Hatch et al. (1989, appendix 3) list a subset of the Hatch et al. (1984) data (72 samples from their locations 8 to 18 in southwestern Missouri, southeastern Kansas, and northeastern Oklahoma). Twenty-seven of these samples (offshore shale lithofacies) have a potential to generate hydrocarbons. Organic-carbon content for these samples ranges from 3.7 to 21.6%, genetic potential, 4.2 to 83 mg/g, hydrogen indices, 100 to 400 mg/g, T<sub>max</sub>, 438° to 452°C, and pristane/phytane, 1.4 to 4.5 (*n* = 17). The T<sub>max</sub> range indicates that organic matter in these samples is marginally mature to mature with respect to petroleum generation.
- (11) Wenger et al. (1988, table 1) show that for 11 samples of core from the 22.7-inch (57.6-cm)-thick Excello shale member of the Fort Scott Limestone from Greenwood County, Kansas, organic-carbon contents range from 0.97 to 25.2%; hydrogen indices, 97 to 360 mg/g;  $\delta^{13}C_{sat}$ , 26.8 to 27.50/oo; and  $\delta^{13}C_{arom}$ , -26.8 to 27.80/oo. They also show that for 19 samples of core from the 36.7-inch (93.3-cm)-thick Middle Pennsylvanian Little Osage Shale Member of the Fort Scott Limestone from Rogers County, Oklahoma, organic-carbon contents range from 1.8 to 20.1%; hydrogen indices, 69 to 410 mg/g;  $\delta^{13}C_{sat}$ , -27.9 to 28.90/oo; and  $\delta^{13}C_{arom}$ , -26.5 to 28.70/oo.  $T_{max}$  for these samples was not reported.
- (12) Ece (1989, table 3) lists organic-carbon contents and hydrogen indices for nine samples of the Excello shale member of the Fort Scott Limestone collected from outcrops in southeastern Kansas and northeastern Oklahoma. For these samples, organic-carbon contents range from 1.9 to 13.4%; hydrogen indices, 69 to 420 mg/g; and T<sub>max</sub>, 424° to 427°C. This T<sub>max</sub> range shows that organic matter in these samples is immature with respect to petroleum generation.
- (13) Hatch and Leventhal (1992) show that for 21 core samples from the 20.8-inch (52.8-cm)-thick Upper Pennsylvanian Stark Shale Member of the Dennis Limestone in Wabaunsee County, Kansas, organic-carbon contents range from 0.5 to 29.5%, hydrogen indices from 55 to 460 mg/g, and T<sub>max</sub>, 420° to 432°C. This T<sub>max</sub> range shows that organic matter in these samples is immature to marginally mature with respect to petroleum generation.
- (14) Hatch and Leventhal (1997) in their study of geochemical processes that affect organic-matter compositions and sulfides in the Excello shale member of the Fort Scott Formation, list organic-carbon contents, hydrogen indices,  $T_{\text{max}}$ , and organic matter  $\delta^{13}$ C for 44 samples of three Middle Pennsylvanian marine black shales from southern Iowa, Missouri, southeastern Kansas, and northeastern Oklahoma.

• (15) Newell et al. (1987) modeled the thermal maturation of organic matter in the Paleozoic section in the Forest City basin utilizing the time-temperature index calculations of Waples (1980). They showed that the Ordovician strata along the axis of the Forest City basin can be in the initial stages of oil generation. Further east in the Easton-McLouth production areas, Ordovician strata can also be in the initial stages of oil generation, but higher geothermal gradients would have to have occurred to compensate for lesser present and past burial (Newell et al., 1987). Maturation sufficient to generate oil in strata

younger than Ordovician is difficult to achieve using present heat flows and reasonable paleoburial depths. More detailed modeling of paleotemperatures and organic matter maturation on the Cherokee platform farther south by Förster and Merriam (1994) shows similar problems in attempting to account for observed vitrinite reflectance in the Cherokee Group. Here, higher maturation can be achieved by invoking flow of heated waters out of the deep Anadarko and Arkoma basins (Förster and Merriam, 1994). The extent to which this fluid flow would affect the Forest City basin is unknown.

## Samples

This study reports geochemical data for 26 oils and 123 samples of potential hydrocarbon source rocks. Locations and descriptive information for the oil samples are in appendix 1; sampled locations are shown in fig. 2A. Four oils (index map locations C through F) are from the

broad belt of oil and gas fields that extends southsouthwestward from the Kansas City region on the eastern flank of the basin; two oils (index map locations O and R) are from fields (Easton and Vassar) to the west and downdip of the first trend. Eighteen oils (index map

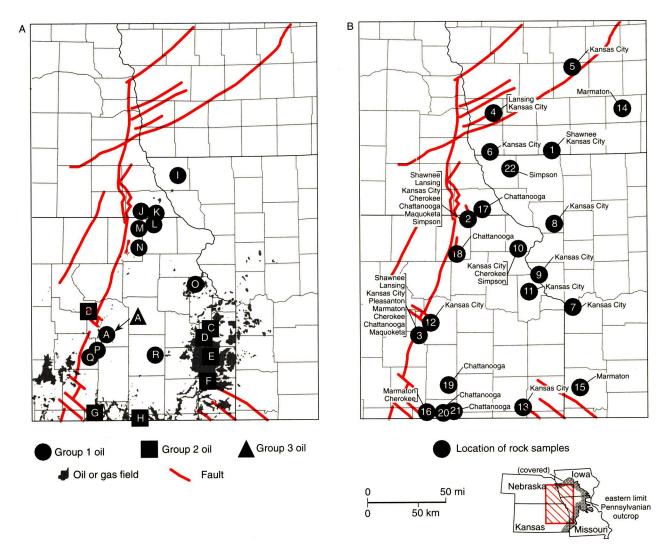


FIGURE 2—(A) OIL-SAMPLE LOCATIONS (A–R) AND (B) ROCK-SAMPLE LOCATIONS (1–22) FROM THE FOREST CITY BASIN AREA IN EASTERN KANSAS, NORTHWESTERN MISSOURI, SOUTHWESTERN IOWA, AND SOUTHEASTERN NEBRASKA. Oil-sample locations are listed in appendix 1; rock-sample locations, appendix 2. Oil Groups 1-3 are defined in the text. Formation or group sampled at each location is shown in B.

locations A, I–N, P, and Q) are from fields found along the north-northeast–south-southwest-trending axis of the basin. For comparison with Forest City basin oils, three additional oils (index map locations B, G, and H) are included in appendix 1. Two of these oils (locations G and H) were collected from fields in Greenwood and Woodson counties, Kansas, on the northern edge of the Cherokee platform region. The third oil (location B) was collected from the Yaege field, Riley County, Kansas, on the Nemaha uplift.

Locations and descriptive information for the 123 samples of potential hydrocarbon-source rocks are in appendix 2; sampled locations are shown in fig. 2B. These samples were collected from conventional and sidewall cores, strip mines, and well cuttings at 21 locations in southwestern Iowa, southeastern Nebraska, northwestern Missouri, and eastern Kansas. Twenty-two samples are from shales and limestones in the Middle Ordovician Simpson Group; three samples, Upper Ordovician Maquoketa Shale; 20 samples, Upper Devonian-Lower Mississippian Chattanooga Shale; 22 samples, Middle Pennsylvanian (Desmoinesian) Cherokee and Marmaton Groups; and 56 samples, Upper Pennsylvanian (Missourian and Virgilian) Pleasanton, Kansas City, Lansing, and Shawnee Groups. The samples from the Middle and Upper Pennsylvanian were primarily collected from thin, widespread, marine black shales that represent the offshore shale lithofacies (Heckel, 1977, 1991; Ebanks et al., 1979).

Descriptions for many of the oil and rock samples included in this report have been previously published. Descriptions for oils from locations G and H are from Hatch et al. (1989, appendix 1); the 10 oils from locations A. K-N. P. and O are from Hatch et al. (1987, table 1). Descriptions for the samples from the Stark Shale Member of the Dennis Limestone from location 12 are from Hatch and Leventhal (1992, table 1); for the Cherokee Group samples from locations 15 and 16, Hatch et al. (1984, table 1); and for the Chattanooga Shale samples from locations 12, 20, and 21, Hatch et al. (1989, appendices 1 and 2). Descriptions for the samples of the Upper Pennsylvanian shales from locations 1, 4-6, and 14 are from core logs at the Iowa Geological Survey. Descriptions for samples of the Stark Shale Member of the Dennis Limestone from locations 11 and 13 are from core logs at the Kansas Geological Survey, whereas descriptions for samples of the Upper Pennsylvanian shales from location 7 were provided by R. M. Coveney, University of Missouri-Kansas City.

Available core samples of the Chattanooga Shale in eastern Kansas represent only a fraction of the thickness of the formation. As examples, the samples of Chattanooga Shale from location 12 (Davis A #2 well) represent the basal 3 ft (0.9 m) of the formation, for location 20 (Bock #1 well) the basal 12 ft (3.7 m) of shale, and for location 21 (Stauffer #1 well) the basal 9 ft (2.7 m) of shale. In this part of eastern Kansas, the Chattanooga Shale is between 50 and 100 ft (15–30 m) thick (Adler et al., 1971, fig. 13).

## **Analytical Methods**

In this report, organic-carbon content (weight %) hydrogen index (HI, mg HC/g TOC), oxygen index (OI, mg CO<sub>2</sub>/g TOC), T<sub>max</sub> (°C), genetic potential (S<sub>1</sub>+S<sub>2</sub>, mg HC/g rock) and production index (PI, S<sub>1</sub>/[S<sub>1</sub>+S<sub>2</sub>]) from Rock-Eval® pyrolysis are used to characterize organic-matter amount, composition, thermal maturity, hydrocarbon-generation potential, and the conversion of organic matter to hydrocarbons. These measurements were determined by the Delsi–Nermag model II instrument with the attached carbon analyzer. Methods and interpretations of the Rock-Eval® results follow Espitalié et al. (1977), Tissot and Welte (1978, p. 445–447), Orr (1983), and Peters (1986).

Oil samples were separated into saturated-hydrocarbon, aromatic-hydrocarbon, resin (NSO), and asphaltene fractions by a Baker–10 Extraction System (J. T. Baker Chemical Co.) using SPE (3-ml) 20-micron filtration columns, silica-gel columns (6 ml), and cyano (CN) columns (6 ml). Elutants used were cyclohexane, cyclohexane/benzene (3/1), and chloroform after Anders et al. (1987). Powdered rock samples were extracted by chloroform (CHCl<sub>3</sub>) in a Soxhlet apparatus for 24 hours. Sulfur was removed by refluxing with polished copper strips. The extract solution was filtered, concentrated

under a stream of nitrogen at room temperature, and then diluted with *n*-heptane to precipitate asphaltenes. A concentrate of the solution was separated by column chromatography on silica gel, eluting successively with *n*-heptane, benzene, and benzene-methanol (1:1 v/v) to collect the saturated-hydrocarbon, aromatic-hydrocarbon, and resin (NSO) fractions, respectively.

Gas chromatography of the saturated-hydrocarbon fractions of both oils and rock extracts was performed by a Hewlett-Packard 5880 gas chromatograph equipped with fused-silica capillary column (50 m × 0.32 mm I.D., SE-54). Oven temperature was programmed from 50° to 320°C at 4°C min <sup>-1</sup>. Detector output was digitized and stored on computer disk for calculation of relative amounts of hydrocarbon components. Absolute quantification of peaks was not attempted. Gas chromatography-mass spectrometry (GC-MS) was performed using a HP 5890 gas chromatograph interfaced to a VG 7035 doublefocusing mass spectrometer. Relative amounts of biomarker compounds were determined by single-ion monitoring of characteristic ions. Relative peak heights of m/z = 191 fragmentograms were used to compare distributions of terpane compounds. The identified terpane compounds are listed in table 1.

TABLE 1—LIST OF IDENTIFIED TERPANES.

Peak	Compound Name
1.	C <sub>19</sub> tricyclic
2.	C <sub>21</sub> tricyclic
3.	C <sub>23</sub> tricyclic
4.	C <sub>25</sub> tricyclic
5.	C <sub>26</sub> tricyclics
6.	C <sub>24</sub> tetracyclic
7.	C <sub>28</sub> tricyclics
8.	C <sub>29</sub> tricyclics
9.	C <sub>29</sub> norhopane
10.	$C_{30}$ -17 $\alpha(H)$ , 21 $\beta(H)$ hopane
11.	$C_{31}$ -17 $\alpha$ (H), 21 $\beta$ (H) homohopane (s + r)
12.	$C_{32}$ -17 $\alpha$ (H), 21 $\beta$ (H) bishomohopane (s + r)
13.	$C_{33}$ -17 $\alpha$ (H), 21 $\beta$ (H) trishomohopane (s + r)

To determine carbon-isotope composition, saturatedand aromatic-hydrocarbon fractions from oils and rock extracts were combusted to carbon dioxide in a highvacuum furnace and gas-transfer system. Stable-carbon isotope ratios were determined with a Finnigan MAT 251, isotope-ratio mass spectrometer. Isotope compositions of carbon dioxide from the samples were compared directly to reference standards of carbon dioxide prepared from NBS-19 limestone ( $\delta^{13}$ C = 1.96% PDB). Ratios are reported as standard per mil (%)00) deviation relative to the Peedee belemnite standard (PDB):

$$\delta^{_{13}}C = [(R \ sample/R \ standard) - 1] \times 10^3,$$
 where: 
$$R = ratio \ of \ ^{_{13}}C \ to \ ^{_{12}}C.$$

## Results

#### **Oils**

The saturated-hydrocarbon fractions from the oils show three gas chromatographic patterns (figs. 3A-C). The first pattern (Group 1, fig. 3A) is shown by 18 oils and is characterized by a dominance of odd-carbon-number nalkanes between  $nC_{10}$  and  $nC_{20}$ , relatively small amounts of branched and cyclic alkanes (including the isoprenoids), and relatively small amounts of alkanes with carbon numbers greater than  $nC_{19}$ . The second pattern (Group 2, fig. 3B), shown by seven oils, is characterized by decreasing amounts of n-alkanes, and the presence of isoprenoids (e.g., pristane and phytane). The third pattern (Group 3, fig. 3C) shown by one oil, is characterized by isoprenoid compounds dominant over the n-alkanes, with maximum n-alkane content at  $nC_{15}$ .

Pristane/phytane, pristane/nC<sub>17</sub>, and carbon-preference index (CPI, after Bray and Evans, 1961) for n-alkanes between  $nC_{12}$  and  $nC_{20}$  for the oils are listed in table 2. Median values and ranges for these parameters are summarized in table 4. A plot of oil pristane/nC<sub>17</sub> versus CPI for  $nC_{12}$ - $nC_{20}$  is shown in fig. 4A. Figure 4A shows clear differences in pristane/ $nC_{17}$  to  $nC_{12}$ - $nC_{20}$  CPI for the three oil groups. For the Group 1 oils, pristane/nC<sub>17</sub> ranges from 0.04 to 0.10; Group 2 oils, 0.49 to 0.67; and for the Group 3 oil, 3.1. The  $nC_{12}$ - $nC_{20}$  CPI for the Group 1 oils is higher (range = 1.5 to 1.6) than for the Group 2 oils (range = 1.0 to 1.1) or the Group 3 oil (1.1).

The distributions and relative amounts of tricyclic, C<sub>24</sub> tetracyclic, and pentacyclic terpanes (m/z = 191) for the oils are shown in figs. 5A-C. Terpane distributions for Group 1 oils (e.g., fig. 5A) are characterized by relatively abundant pentacyclic terpanes (peaks 9-13), minimal amounts of the tricyclic terpanes, and intermediate amounts of C<sub>24</sub> tetracyclic terpane (peak 6). Terpane

distributions for the Group 2 oils (fig. 5B) and the Group 3 oil (fig. 5C) are similar to each other, with hopane (C<sub>30</sub>, peak 10) the most abundant pentacyclic terpane, C23 (peak 3) the most abundant tricyclic terpane, relatively abundant extended tricyclic compounds (C26, C28, and C29 doublets, peaks 5, 7, and 8), and low relative amounts of C<sub>24</sub> tetracyclic terpane. Relative to amounts of pentacyclic terpanes, the amounts of tricyclic terpanes are relatively greater in the Group 2 oils compared to the Group 3 oil.

Measures of whole oil  $\delta^{13}$ C for 10 oils and  $\delta^{13}$ C of the saturated- and aromatic-hydrocarbon fractions for 15 oils are listed in table 2. Whole oil  $\delta^{13}$ C for six Group 1 oils ranges from -26.7 to -31.3%, a 4.6% difference; for the three Group 2 oils, the range is -29.9 to -30.7% oo, a  $0.8^{\circ}/_{oo}$  difference. Oil  $\delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  measures show similar trends. For 11 Group 1 oils,  $\delta^{13}C_{sat}$  ranges from -26.8 to -31.3%, a 4.5% difference; for three Group 2 oils,  $\delta^{13}$ C<sub>sat</sub> ranges from -29.6 to -30.50/oo, a 0.90/oo difference;  $\delta^{13}C_{arom}$  for the Group 1 oils ranges from -26.2 to -31.1, a difference of 4.90/oo;  $\delta^{13}C_{arom}$  for the Group 2 oils ranges from -28.9 to -29.4%, a difference of 0.5%.

## **Hydrocarbon Source-rock Evaluation**

Hydrocarbon source-rock potential is dependent on the amount, thermal maturity, and composition of organic matter in the rock. The amount of organic matter is measured by the organic carbon content [TOC]; relative organic matter compositions, by hydrogen index [HI] and oxygen index [OI]. The relative potential to generate hydrocarbons is measured by genetic potential (GP), thermal maturity by  $T_{max}$  (°C), and vitrinite reflectance (R<sub>o</sub>%), whereas, the conversion of organic matter to hydrocarbons is indicated by production index (PI). TOC, HI, OI, GP,  $T_{max}$ , and PI for 87 core and mine samples are

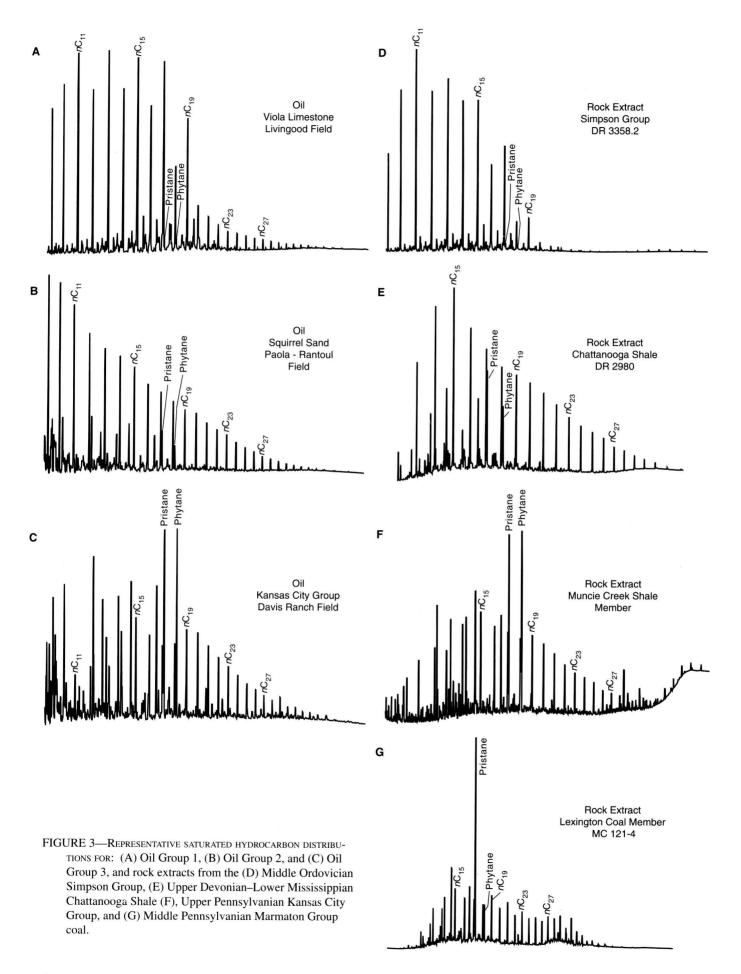


TABLE 2—Organic Geochemical Characteristics of 26 oil samples from Northwestern Missouri, southeastern Nebraska, and EASTERN KANSAS. (Oil groups are defined in the text; oil-sample locations are shown in fig. 2A.)

Location map letter	Sample name	°API	Pristane/ Phytane <sup>a</sup>	Pristane/ nC <sub>17</sub> <sup>a</sup>	nC <sub>12-20</sub> b CPI	$\delta^{13}C_{oil}{}^{c}$	δ <sup>13</sup> C <sub>sat</sub> <sup>c</sup>	δ <sup>13</sup> C <sub>arom</sub>
Group 3								
A	Davis A#15	30	0.9	3.1	1.1	-29.5	-30.1	-29.1
Group 2								
В	Yaege #12	23	1.6	0.57	1.1	-29.9	-30.5	-29.3
C	Keegan	31	1.6	0.56	1.0	-30.7		
D	Kramer	20	1.4	0.67	1.0	-29.9		
E	Andersen		1.6	0.53	1.0			
F	Polhamus		1.5	0.56	1.0			
G	Teeter #37		1.6	0.52	1.0		-30.2	-29.4
Н	Headley A		1.6	0.49	1.1		-29.6	-28.9
Group 1								
I	#1 Watkins		1.0	0.10				
J	#1 Miles A		0.9	0.09	1.6			
K	Reesman		0.9	0.08	1.5		-29.0	-28.7
K	Seiberson		0.9	0.07	1.5			
K	Sand Rock		1.0	0.07	1.5		-28.8	-29.0
K	Harper #7	31	0.9	0.08	1.5	-29.0	-29.1	-28.9
K	Walldraff		0.9	0.08	1.5			
K	Liniger		0.9	0.07	1.6			
L	Hustead		1.0	0.07	1.5		-29.7	-29.7
M	#1 Aberle	25	0.8	0.08	1.6	-30.6	-30.4	-30.6
M	#1 Parli		0.8	0.06	1.6		-30.0	-30.2
N	#1 Potts-V	29	0.8	0.06	1.6	-29.3	-29.3	-29.0
N	#1 Potts-S	28	0.8	0.04	1.6	-31.3	-31.3	-31.1
0	Crook Lease	23	1.2	0.07	1.5	-28.9		
A	Davis A#11	28	0.9	0.09	1.5	-26.7	-26.8	-26.2
P	#8 Olsen		0.7	0.07	1.5		-26.9	-26.6
Q	#1 Greer		0.8	0.06	1.6		-27.1	-26.8
R	#1 Woodard		1.2	0.05	1.6			

Blank space indicates no data.

Sample locations are listed in appendix 1.

a. Ratios calculated from relative peak heights above base line.

b.  $\frac{nC_{12-20}}{\text{CPI}} = \frac{1}{2} \left[ \frac{nC_{13} + nC_{15} + nC_{17} + nC_{19}}{nC_{12} + nC_{14} + nC_{16} + nC_{18}} + \frac{nC_{13} + nC_{15} + nC_{17} + nC_{19}}{nC_{14} + nC_{16} + nC_{18} + nC_{20}} \right]$ Modified from Bray and Evans (1961).

c. In parts-per-mil relative to the PDB marine-carbonate standard.

 $<sup>\</sup>delta^{\scriptscriptstyle{13}}C_{\scriptscriptstyle{oil}}$  values were provided by Sohio Petroleum Geochemistry Group.

 $<sup>\</sup>delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  for the Teeter #37 and Headley A #1-9 oils are from Hatch et al. (1989, table 5).

 $<sup>\</sup>delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  for the Group 1 oils except #7 Harper are from Hatch et al. (1987, table 2).

 $<sup>\</sup>delta^{\scriptscriptstyle 13}C_{sat}$  and  $\delta^{\scriptscriptstyle 13}C_{arom}$  for the #7 Harper and Yaege #12 oils were provided by Sohio Petroleum Geochemistry Group.

TABLE 3—Organic geochemical characteristics of bitumens extracted from 24 samples of Paleozoic rocks from northwestern Missouri and northeastern Kansas. The Kansas City Group is of Upper Pennsylvanian age. The Marmaton Group and Cherokee Group are of Middle Pennsylvanian age; the Chattanooga Shale is of Upper Devonian—Lower Mississippian age; whereas, the Simpson Group is of Middle Ordovician age.

Location map number	Sample number	Bitumen mg/kg <sup>a</sup>	Pristane/ Phytane <sup>b</sup>	Pristane/ nC <sub>17</sub> b	nC <sub>12-20</sub> <sup>c</sup> CPI	$\delta$ $^{13}C_{sat}$ $^d$	$\delta^{13}C_{arom}^{d}$
Kansas Cit	y Group		200				
7	RC12	5700	1.0	2.2	1.1	-29.4	-29.2
	RC5	6300	1.1	1.1	1.0	-28.9	-29.0
12	S2-1	7050	1.3	1.1	1.1		
	S4-1	5600	1.4	0.6	1.1		
	S5	5590	1.1	1.3	*		
	S8T	6360	1.1	2.0	1.1		
	S11	9840	1.2	2.8	1.1		
	S12B	5050	1.4	0.8	1.1		
13	W5A6	5750	1.3	1.8	1.1	-28.7	-28.4
	W5A12	2930	1.4	0.8	1.1	-28.3	-27.9
	W5A16	2310	1.2	1.8	1.1	-28.5	-28.1
Marmaton	Group						
15	MC121-2	6700	1.4	3.2	*		
	MC121-3	2000	1.8	2.7	*		
	MC121-4	9400	6.7	4.7	*	-26.6	-24.9
16	349-3	6300	1.7	1.3	*	-27.4	-27.6
Cherokee (	Group						
16	349-46	3800	1.8	1.1	*		
	349-57	2800	2.0	1.9	*		
Chattanoo	ga Shale						
12	DR2980	4000	1.6	0.83	*	-29.1	-28.4
20	STA2319	5300	1.6	0.55	*	-29.3	-28.7
21	BOC2174.8	2094	1.5	0.50	1.1		
Simpson G	roup						
12	DR3313.5	3800	1.5	0.07	1.4	-26.3	-26.7
	DR3315.5	1010	0.7	0.06	*	-26.4	-26.1
	DR3321.0	580	0.8	0.05	*	-25.6	-25.4
	DR3358.2	4240	1.5	0.07	1.4		

Blank space indicates no data.

Sample locations are listed in appendix 2.

Modified from Bray and Evans (1961); \*indicates CPI could not be calculated because some of the  $nC_{12}$  to  $nC_{20}$  compounds were lost during evaporation of excess solvent.

a. gHC/kg sample

b. Ratios calculated from relative peak heights above base line.

c.  $nC_{12-20} = \frac{1}{2} \left[ \frac{nC_{13} + nC_{15} + nC_{17} + nC_{19}}{nC_{12} + nC_{14} + nC_{16} + nC_{18}} + \frac{nC_{13} + nC_{15} + nC_{17} + nC_{19}}{nC_{14} + nC_{16} + nC_{18} + nC_{20}} \right]$ 

d. In parts-per-mil relative to the PDB marine-carbonate standard.

Total bitumen and pristane/phytane ratios for Marmaton Group, Cherokee Group, and Chattanooga Shale samples are from Hatch et al. (1984, table 7; 1989, table 5 and appendix 3).

 $<sup>\</sup>delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  for the Simpson Group samples are from Hatch et al. (1987, table 2).

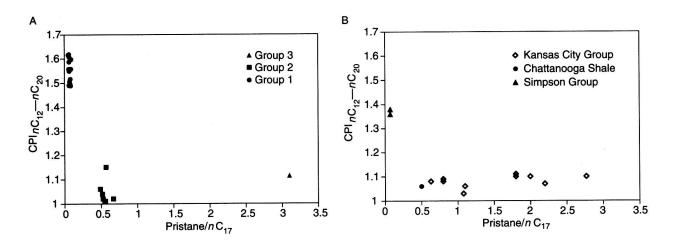
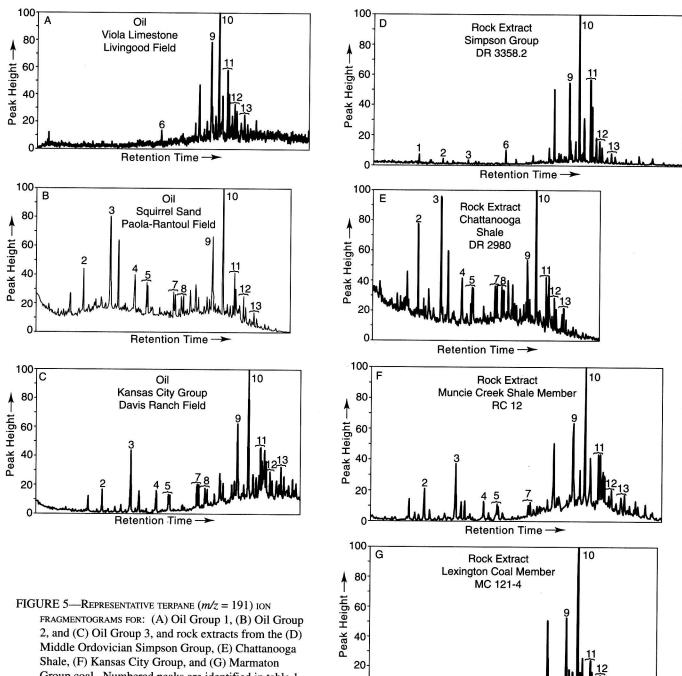


FIGURE 4—Plot of pristane/ $nC_{17}$  versus  $nC_{12}$ - $nC_{20}$  carbon preference index (CPI) for (A) Oil Groups 1, 2, and 3, and (B) rock extracts from shales of the Middle Ordovician Simpson Group shales, Upper Devonian-Lower Mississippian Chattanooga Shale, and Upper Pennsylvanian Kansas City Group.

TABLE 4—Median values and ranges of pristane/phytane, pristane/ $nC_{17}$ ,  $nC_{12}$ - $nC_{20}$  CPI for Oil Groups 1, 2, and 3, and rock ex-TRACTS FROM SHALES in the Upper Pennsylvanian Kansas City Group and Middle Pennsylvanian Marmaton and Cherokee Groups, Upper Devonian-Lower Mississippian Chattanooga Shale, and Middle Ordovician Simpson Group shales from the Forest City basin and adjacent areas. (Data are listed in tables 2 and 3.)

	Pristane/phytane	Pristane/nC <sub>17</sub>	$nC_{12}$ - $nC_{20}$ CPI
Group 3 oil	0.9	3.1	1.1
oup 2 oils	1.6 (n = 7) $1.4-1.6$	0.56 (n = 7) 0.49-0.67	1.0 ( <i>n</i> = 7) 1.0-1.1
oup 1 oils	0.9 (n = 18) 0.7-1.2	0.07 (n = 18) 0.04-0.10	1.6 (n = 17) 1.5-1.6
nsas City Group	1.2 ( <i>n</i> = 11) 1.0-1.4	1.3 ( <i>n</i> = 11) 0.6-2.8	1.1 ( <i>n</i> = 10) 1.0-1.1
maton and Cheroke oups	e $1.8 (n = 5)$ 1.4-2.0	2.7 (n = 11) 1.1-3.2	*
attanooga Shale	1.6 ( <i>n</i> = 3) 1.5-1.6	0.55 (n = 3) $0.50 - 0.83$	$1.1\ (n=1)$
npson Group	1.2 (n = 4) $0.7-1.5$	0.06 (n = 4) 0.05-0.07	1.4 (n = 2) $1.36-1.38$

<sup>\* =</sup> not calculated.



**Retention Time** 

Group coal. Numbered peaks are identified in table 1.

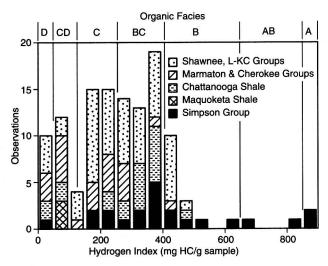


FIGURE 6—STACKED BAR HISTOGRAM OF HYDROGEN INDICES for samples from the Middle Ordovician Simpson Group, Upper Ordovician Maquoketa Shale, Upper Devonian—Lower Mississippian Chattanooga Shale, Middle Pennsylvanian Cherokee and Marmaton Groups, and Upper Pennsylvanian Kansas City (KC), Lansing (L), and Shawnee Groups from the Forest City basin and adjacent areas. Boundaries for organic facies D through A are from Jones (1987) and are discussed in the text.

listed in appendix 3; similar data for 36 cuttings samples, appendix 4. Vitrinite reflectance for 17 cuttings samples are included in appendix 4. Median values and ranges for TOC, HI, and GP are listed in table 5.

#### **Organic-carbon Content**

Median values and ranges of organic-carbon contents (table 5) show considerable variation for the nine different formations/groups summarized. Median TOC contents are high for shale samples from the Shawnee Group (10.1%), Lansing Group (10.0%), Kansas City Group (12.1%), and Marmaton Group (9.9%); intermediate TOC contents characterize shale samples from the Pleasanton Group (4.4%–8.0%), Cherokee Group (5.0%), and Chattanooga Shale (3.6%). Relatively low TOC contents characterize shale samples from the Maquoketa Shale (1.5%) and shales and limestones from the Simpson Group (1.1%). TOC for the coal samples are highest: 47% TOC for the Cherokee Group coal sample and 61% TOC for the Marmaton Group coal sample.

#### **Organic-matter Composition**

Median values and ranges for hydrogen indices for the nine different formations/groups are listed in table 5. The distribution of hydrogen indices for each formation/group is shown in fig. 6. Median hydrogen index is highest (between 370 and 380 mg/g) for the Simpson Group shale

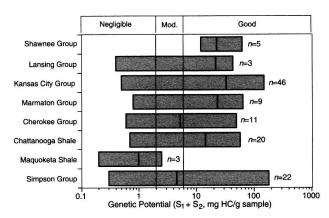


FIGURE 7—Median values and ranges of Genetic Potential for samples from the Middle Ordovician Simpson Group, Upper Ordovician Maquoketa Shale, Upper Devonian–Lower Mississippian Chattanooga Shale, Middle Pennsylvanian Cherokee and Marmaton Groups, and Upper Pennsylvanian Kansas City, Lansing, and Shawnee Groups from the Forest City basin and adjacent areas. Number of observations = n.

and limestone samples and the Chattanooga Shale samples (350 mg/g). Lower median values are shown for samples of Kansas City Group shales (295 mg/g), Marmaton and Cherokee Group coals (230 mg/g), Shawnee Group and Marmaton Group shales (220 mg/g), and Cherokee Group shales (170 mg/g). The lowest median values were obtained for the Lansing Group shales (130 mg/g), Pleasanton Group shales (between 120 and 140 mg/g), and Maquoketa Shale samples (70 mg/g). The range of hydrogen index measures for each formation/group can be considerable. As examples, hydrogen index for the Kansas City Group samples ranges from 13 to 505 mg/g; Chattanooga Shale samples, 210 to 490 mg/g; and Simpson Group samples, 11 to 1,160 mg/g.

#### **Hydrocarbon Source-rock Potential**

Ranges and median values of genetic potential for samples from each formation/group are listed in table 5 and shown in fig. 7. Median genetic potential is highest for shale samples from the Kansas City Group (32 mg/g), Chattanooga Shale (25 mg/g), Marmaton Group (23 mg/g), Shawnee Group (22 mg/g), and Lansing Group (21 mg/g). Lower median values were measured for shale samples from the Pleasanton Group (between 5.7 and 12 mg/g), Cherokee Group (5.0 mg/g), and Simpson Group shales and limestones (4.0 mg/g). The lowest median value was for the Maquoketa Shale (1.0 mg/g). As was the case for hydrogen index, the range of genetic potential for each formation/group can be considerable. As examples, genetic potential for the Kansas City Group samples ranges from 0.5 to 150 mg/g, Chattanooga Shale samples, 8 to 57 mg/g, and Simpson Group samples, 0.3 to 180 mg/g.

TABLE 5—Median values of organic carbon, hydrogen index, and genetic potential for core, sidewall core, and well cuttings of shale samples from the Upper Pennsylvanian Shawnee Group, Lansing Group, and Kansas City Group, Middle Pennsylvanian Marmaton Group, Cherokee Group, Upper Devonian–Lower Mississippian Chattanooga Shale, Upper Ordovician Maquoketa Shale, and shale and limestone samples from the Middle Ordovician Simpson Group in the Forest City basin and adjacent areas. (Data are listed in appendices 3 and 4.)

	Organic carbon (weight %)	Hydrogen Index (mg/g) <sup>a</sup>	Genetic Potential (mg/g) <sup>b</sup>
Shawnee Group $(n = 5)$	10.1	220	22
	(5.2-21.0)	(205-365)	(12-51)
Lansing Group $(n = 3)$	10.0	130	21
	(2.0-15.2)	(19-415)	(0.43-43)
Kansas City Group $(n = 46)$	12.1	295	32
	(0.9-30.8)	(13-505)	(0.5-150)
Pleasanton Group $(n = 2)$	*	*	*
• • •	(4.4-8.0)	(120-140)	(5.7-12)
Marmaton Group $(n = 9)$	9.9	220	23
•	(1.1-21.6)	(38-400)	(0.8-64)
Cherokee Group $(n = 11)$	5.0	170	6.1
• • •	(1.1-14.9)	(26-410)	(0.6-49)
Chattanooga Shale $(n = 16)$	3.6	350	25
	(1.9-4.4)	(210-490)	(8-57)
Maquoketa Shale $(n = 3)$	1.5	70	1.0
• ,	(0.3-3.1)	(60-90)	(0.2-2.3)
Simpson Group $(n = 22)$	1.1	370-380	4.0
• • • • • •	(0.3-15.8)	(11-1,150)	(0.3-180)

HC = hydrocarbons, TOC = total organic carbon.

n =number of samples.

Summaries for the Chattanooga Shale are based on cores only.

## **Organic-matter Thermal Maturity**

For all stratigraphic units, T<sub>max</sub> of organic matter generally increases from north to south.  $T_{max}$  ranges from 411° to 433°C for shales from the Kansas City, Lansing, and Shawnee Groups from southwestern Iowa, northwestern Missouri, and Nemaha and Doniphan counties, Kansas (locations 1, 2, and 4-11, figs. 2B and 8C). For the Upper Pennsylvanian rocks in the southwestern and southern parts of the basin, T<sub>max</sub> is more variable, ranging from 423° to 439°C at locations 12 and 13 in Wabaunsee and Bourbon counties, Kansas, to 434° to 444°C at location 3 in southwestern Wabaunsee County (figs. 2B and 8C).  $T_{max}$  ranges from 421° to 436°C for organic matter in the Cherokee and Marmaton Groups at location 14 in Clarke County, Iowa, and locations 2 and 10 in Nemaha and Doniphan counties in northeastern Kansas (figs. 2B and 8C). For the Middle Pennsylvanian rocks in the southwestern and southern parts of the basin, T<sub>max</sub> ranges from 436° to 460°C for cuttings samples from location 3 in southwestern Wabaunsee County, Kansas, and location 15 in Bates County, Missouri, and for core

samples from location 16 in Greenwood County, Kansas, on the northern edge of the Cherokee platform (figs. 2B and 8C).

A similar general increase in  $T_{max}$  from north to south is characteristic of organic matter in the Chattanooga Shale (figs. 2B and 8B) and in the Simpson Group (figs. 2B and 8A). For the Chattanooga Shale, T<sub>max</sub> ranges from 431° to 438°C at location 17 in Richardson County, Nebraska; location 2 in Nemaha County, Kansas; and at location 12 in Wabaunsee County, Kansas. T<sub>max</sub> ranges from 440° to 445°C in a sample of cuttings from location 3 in Wabaunsee County, Kansas, and in core samples from locations 19-21 in Lyons and Greenwood counties, Kansas. For the Simpson Group, T<sub>max</sub> of organic matter ranges from 436° to 444°C at location 22 in Atchison County, Missouri; from 441° to 446°C at location 2 and 10 in Nemaha and Doniphan counties, Kansas; and from 433° to 447°C for Simpson Group rocks at location 12 in Wabaunsee County, Kansas.

Vitrinite-reflectance ( $R_o$ ) measures (appendix 4) range from 0.39%? to 0.59% for six samples from the Kansas City, Lansing, and Shawnee Groups from locations 2 and 10 in Nemaha and Doniphan counties, Kansas. The range

 $a = S_2/TOC$ , mg HC/g TOC.

 $b = S_1 + S_2$ , mg HC/g TOC.

<sup>\* =</sup> value not calculated.

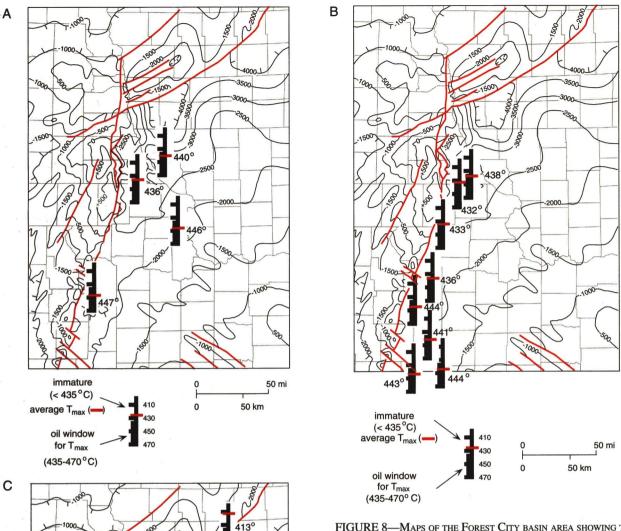


FIGURE 8—Maps of the Forest City basin area showing the Mean Rock-Eval® T<sub>max</sub> for organic matter in (A) limestones and shales from the Middle Ordovician Simpson Group and Upper Ordovician Maquoketa Shale, (B) Upper Devonian—Lower Mississippian Chattanooga Shale, and (C) shales from the Middle Pennsylvanian Cherokee and Marmaton Groups and Upper Pennsylvanian Pleasanton, Kansas City, Lansing, and Shawnee Groups. T<sub>max</sub> measures <435°C are immature with respect to petroleum generation; T<sub>max</sub> measures from 435° to 470°C are mature (Peters, 1986).

is from 0.55% to 0.60% for four samples from location 3 in southwestern Wabaunsee County, Kansas.  $R_o$  ranges from 0.47% to 0.56% for three Cherokee Group samples from locations 2 and 10 in Nemaha and Doniphan counties, Kansas; for the one sample from location 3 in Wabaunsee County, Kansas,  $R_o$  is 0.70%. For two Chattanooga Shale samples from locations 2 and 18 in Nemaha County, Kansas,  $R_o$  is 0.58% and 0.63%; for the one Chattanooga Shale sample at location 3,  $R_o$  is 0.60%, which is less than the  $R_o$  (0.70%) of the Cherokee Group sample 200 ft (61 m) higher in the section.

423°

#### **Hydrocarbon Generation**

For the Forest City basin in Iowa, Missouri, and northeastern Kansas, production indices (for samples with  $\geq 1.0\%$  TOC) are  $\leq 0.10$  for 66 of 69 samples from the Middle and Upper Pennsylvanian shales for four of 17 Chattanooga Shale samples, for two of three Maquoketa Shale samples, and for 13 of 14 Simpson Group shale and limestone samples.

#### **Rock-extract Compositions**

Saturated hydrocarbon distributions from extracts of potential hydrocarbon-source rocks from the Forest City basin follow four chromatographic patterns (figs. 3D-G). The first pattern (fig. 3D), typical of four shale samples in the Simpson Group, is characterized by a dominance of odd-carbon number *n*-alkanes between  $nC_{10}$  and  $nC_{20}$ , relatively small amounts of branched and cyclic alkanes, and relatively small amounts of n-alkanes with carbon numbers greater than  $nC_{19}$ . The second pattern (fig. 3E), typical of three samples of the Chattanooga Shale, is characterized by smoothly decreasing amounts of nalkanes, and the presence of intermediate contents of isoprenoids (e.g., pristane and phytane). The third pattern (fig. 3F), typical of distributions from 16 samples of Middle and Upper Pennsylvanian shales, is primarily characterized by pristane and phytane dominant or equal in abundance to the *n*-alkanes. The fourth pattern (fig. 3G) from the Marmaton Group coal (MC121-4) is characterized by pristane dominant over the other isoprenoids and the *n*-alkanes.

Pristane/phytane, pristane/ $nC_{17}$  and carbon-preference number for the n-alkanes between  $nC_{12}$  and  $nC_{20}$  for the rock extracts are listed in table 3. Median values and ranges are summarized in table 4. Pristane/ $nC_{17}$  versus  $nC_{12}$ - $nC_{20}$  CPI for the rock extracts is illustrated in fig. 4B. Figure 4B shows major differences in pristane/ $nC_{17}$  versus  $nC_{12}$ - $nC_{20}$  CPI. For Simpson Group shale extracts,

pristane/ $nC_{17}$  ranges from 0.05 to 0.07, for Chattanooga Shale extracts, 0.50 to 0.83, for Upper Pennsylvanian shale extracts, 0.60 to 2.8, and for Middle Pennsylvanian shale extracts, 1.1 to 3.2. The  $nC_{12}$ - $nC_{20}$  CPI is higher for the two samples from the Simpson Group (1.4 and 1.4) compared to the Chattanooga Shale (1.1) sample or the seven Upper Pennsylvanian shale samples (1.0 to 1.1).

Measures of  $\delta^{13}C$  of the saturated and aromatic hydrocarbon fractions from extracts from 12 rock samples are listed in table 3.  $\delta^{13}C_{sat}$  for extracts from five Kansas City Group shales and two Chattanooga Shale samples are more negative (range = -28.3 to -29.4°/00) than  $\delta^{13}C_{sat}$  of extracts for the Marmaton Group shale and three Simpson Group samples (range = -25.6 to -26.6°/00). Measures of  $\delta^{13}C_{arom}$  show similar distributions with a  $\delta^{13}C_{arom}$  range of -27.9 to -29.2°/00 for extracts of the Kansas City Group shales and Chattanooga Shale samples, and a  $\delta^{13}C_{arom}$  range of -25.6 to -26.6°/00 for extracts of the Marmaton Group shale and three Simpson Group samples.

The relative amounts of tricyclic (peaks 2–5, 7–8),  $C_{24}$ tetracyclic (peak 6), and pentacyclic (peaks 9–13) terpanes from rock extracts are shown in figs. 5D-G. The terpane distribution in fig. 5D is typical of shales from the Simpson Group and is characterized by minimal amounts of the tricyclic terpanes compared to the pentacyclic terpanes, and C<sub>24</sub> tetracyclic terpane greater than tricyclic terpanes. The terpane distributions from the Chattanooga Shale (fig. 5E) is characterized by equivalent amounts of tricyclic and pentacyclic terpanes, with hopane (C<sub>30</sub>, peak 10) the most abundant pentacyclic terpane,  $C_{23}$  (peak 3) the most abundant tricyclic terpane, relatively abundant extended tricyclic compounds ( $C_{28}$  and  $C_{29}$  doublets, peaks 5 and 7), and relatively low amounts of  $C_{24}$  tetracyclic terpane. The terpane distribution from the Kansas City Group offshore shale (fig. 5F) is similar to that of the Chattanooga Shale except that the pentacyclic terpanes are relatively more abundant than tricyclic terpanes. The terpane distribution from the extract of the Marmaton Group coal (fig. 5G) is similar to those of the Simpson Group shales.

## Discussion

## **Hydrocarbon Source-rock Potential**

For a rock interval to be an effective source of hydrocarbons, a minimum amount of organic matter must be present, and this organic matter must be thermally mature with respect to petroleum generation. The amounts and compositions (oil and/or gas) of generated hydrocarbons are determined by the composition and thermal maturity of the organic matter.

## **Organic-matter Content**

Ronov (1958) investigated some 26,000 samples of different ages and environments from oil and nonoil provinces. This work demonstrated that average organic carbon content was 1.4% for nonreservoir shale-type sediments in petroliferous areas, compared to an average of 0.4% for areas that were nonpetroliferous. Based on this study, Ronov (1958) estimated a critical minimum

organic-carbon content of 0.5% for a shale to be an effective hydrocarbon source rock. Based on this 0.5% organic carbon minimum, the organic-carbon contents listed in appendices 3 and 4 show that most sampled intervals of the Middle and Upper Pennsylvanian offshore shales, Middle Pennsylvanian coals, Chattanooga Shale, Maquoketa Shale, and Simpson Group shales and limestones qualify as potential hydrocarbon source rocks.

## **Organic-matter Composition**

Jones (1987) defines a series of organic facies (A-D) for rocks worldwide based on microscopic characteristics, hydrogen and oxygen indices, and H/C ratios of organic matter that is marginally mature with respect to petroleum generation ( $R_o \sim 0.50\%$ ). Stacked bar histograms of hydrogen indices and organic facies (A-D) for samples from the Forest City basin are shown in fig. 7. In the Jones (1987) classification, organic matter in organic facies A is characterized by HI ≥ 850 mg/g; facies AB, ≅ 850 to 650 mg/g; facies B,  $\cong$  650 to 400 mg/g; facies BC,  $\approx$  400 to 250 mg/g; facies C,  $\approx$  250 to 125 mg/g; facies CD,  $\approx$  125 to 50 mg/g; and facies D,  $\leq$  50 mg/g. Organic matter in organic facies A is either almost entirely recognizable algal remains, often of a single genus or amorphous clumps of a probable combined algal and bacterial origin. Recognizable terrestrial organic matter is negligible. Organic facies AB is usually dominated by highly fluorescent amorphous kerogen of mixed algal and bacterial origin. Terrestrially derived organic matter may be present in detectable amounts. The primary precursor biomass for organic facies B is marine/nonmarine algae and associated bacteria in various stages of decomposition. Terrestrial organic input can be quite variable. Organic facies BC usually is composed of a mixture of partially degraded terrestrial and algal material. Alternatively, this facies can be created by insertion of substantial amounts of terrestrially derived organic matter into an anoxic water column. Organic facies C is usually dominated by terrestrial debris in various stages of oxidation; facies CD is moderate to well-oxidized organic matter, whereas, organic matter in organic facies D is either highly oxidized or redeposited residual organic matter (Jones, 1987).

The volume and type of generated hydrocarbons vary widely with organic facies. Organic facies A is the most oil-generative facies, but is rare in marine shales. Facies AB rocks are prolific sources of oil. Organic facies B is the source of most of the world's oil. Organic facies BC has capacity to generate both oil and gas. Hydrocarbons generated from organic facies C are nearly always condensate and gas; organic facies C coals can generate prodigious amounts of gas. Organic facies CD has a moderate capacity for dry-gas generation, whereas, organic facies D is essentially nongenerative (Jones, 1987).

As summarized in table 5 and shown in fig. 6, organic matter compositions in the Middle and Upper Pennsylvanian shales from the Forest City basin primarily represent organic facies BC, and C, with compositions ranging from facies B through facies D. Organic matter in the Kansas City Group marine shales are more hydrogen rich (median HI = 295 mg/g, organic facies BC), compared to organic matter in the other Pennsylvanian groups (median HI from 220 to 130 mg/g, organic facies C and CD). Where thermally mature, organic matter in the Kansas City Group rocks should generate both oil and gas, whereas, organic matter in the other Pennsylvanian groups should primarily generate gas with some oil. Organic matter in the Chattanooga Shale primarily represents organic facies BC (median HI = 350 mg/g) with the range in compositions from organic facies B to D. Where thermally mature, this organic matter should generate both oil and gas. Organic matter in the Maquoketa Shale falls into organic facies CD and should have little hydrocarbon-generation potential. The Simpson Group shales and limestones have the widest range in organic-matter compositions, from organic facies A to D. Where thermally mature, the intervals containing organic matter with facies A, AB, and B compositions should primarily be source rocks for oil. One additional note: terrestrial organic matter cannot possibly be a component in any sample of the Maquoketa Shale and Simpson Group, as these units predate the evolution of land plants (Tissot and Welte, 1978, p.18).

## **Hydrocarbon-generation Potential**

Genetic potential  $(S_1 + S_2, mg HC/g sample)$  is a summary measure of the relative potential of the rock to generate hydrocarbons and is dependent on the amount, type, and thermal maturity of organic matter. Tissot and Welte (1978, p. 447) suggest the following classification of genetic potential:

- <2 mg/g: little or no source-rock potential,
- $\geq 2$  mg/g and  $\leq 6$  mg/g: moderate source-rock potential,
- >6 mg/g: good source-rock potential.

Median values and ranges of genetic potential listed in table 4 are illustrated in fig. 7. Based on the classification scheme of Tissot and Welte (1978, p. 447), intervals within the Shawnee, Lansing, Kansas City, and Pleasanton Groups, the Cherokee and Marmaton Groups, the Chattanooga Shale, and the Simpson Group all have median genetic potentials of >6 mg/g to >>6 mg/g, indicating good to excellent potential to generate hydrocarbons. The good to excellent hydrocarbon sourcerock potential shown for these stratigraphic intervals in the Forest City basin are consistent with results reported for these same intervals in other areas in the midcontinent region (e.g., Anadarko basin, Burruss and Hatch, 1989; Cherokee platform, Hatch et al., 1989; Iowa shelf, Hatch et al., 1987, and Jacobson et al., 1988) and in the Illinois basin (Hatch et al., 1991).

Median genetic potential for the Maquoketa Shale in the Forest City basin indicates generally poor potential to generate hydrocarbons. This generally poor potential for the Maquoketa Shale is similar to that for most samples of the laterally equivalent Upper Ordovician Sylvan Shale in the Anadarko basin (Burruss and Hatch, 1989, their fig. 3).

#### **Organic-matter Thermal Maturity**

The main zone of oil generation occurs over an R<sub>o</sub> range of about 0.5–0.7% to 1.3%, which corresponds to a  $T_{max}$  of 435° to 460°C (Tissot and Welte, 1978, p. 450– 455).  $T_{max}$  values of 411° to 433°C for shales from the Kansas City, Lansing, and Shawnee Groups from southwestern Iowa, northwestern Missouri, and Nemaha and Doniphan counties, Kansas, show that organic matter in these shales is immature with respect to petroleum generation. T<sub>max</sub> values of 423° to 439°C for Upper Pennsylvanian shales in Wabaunsee and Bourbon counties, Kansas, indicate immature to marginally mature organic matter. In southwestern Wabaunsee County, Kansas,  $T_{max}$ values of 434° to 444°C indicate marginally mature to thermally mature organic matter. T<sub>max</sub> values of 421° to 436°C in shale samples from the Cherokee and Marmaton Groups in Clarke County, Iowa, and in Nemaha and Doniphan counties, Kansas, suggest immature to marginally mature organic matter.  $T_{max}$  values range of 436° to 460°C for samples in southwestern Wabaunsee County, Kansas; Bates County, Missouri; and Greenwood County, Kansas, indicate marginally mature to mature organic matter. This general north-to-south increase (from immature to mature) in organic-matter thermal maturity in the Upper and Middle Pennsylvanian shales is shown in fig. 8C where mean  $T_{max}$  values are plotted.

This general north-to-south increase (from immature to mature) in thermal maturity is also true for organic matter in the Chattanooga Shale (fig. 8B) and the Simpson Group (fig. 8A).  $T_{max}$  measures of 431° to 438°C for the Chattanooga Shale in Richardson County, Nebraska, and Nemaha and Wabaunsee counties, Kansas, suggest immature to marginally mature organic matter, whereas,  $T_{max}$  of 440° to 445°C for the Chattanooga Shale in southwestern Wabaunsee, Lyons, and Greenwood counties, Kansas, suggest thermally mature organic matter.  $T_{max}$  measures of 436° to 444°C for the Simpson Group shales and limestones in Nemaha and Doniphan counties, Kansas,  $T_{max}$  of 441° to 446°C in Atchison County, Missouri, and 433° to 447°C in Wabaunsee County, Kansas, all suggest marginally mature to thermally mature organic matter.

Production indices  $\geq 0.10$  are interpreted to mean that significant hydrocarbon generation has occurred (Espitalié et al., 1977; Tissot and Welte, 1978, p. 445–447). PI is < 0.10 for 63 of 72 samples from the Middle and Upper Pennsylvanian shales, indicating that significant generation of hydrocarbons has not occurred. This conclusion supports previously interpreted  $T_{max}$  values above, where

we concluded that most organic matter in the Middle Upper Pennsylvanian shales is thermally immature to marginally mature with respect to petroleum generation. For eight of the nine Middle and Upper Pennsylvanian shale samples with PI  $\geq$  0.10, organic carbon contents are low, ranging from 0.9 to 2.6%. This relationship suggests that these high relative PI values may represent migrated hydrocarbons. PI ratios  $\geq$  0.10 for 13 of 17 Chattanooga Shale samples suggest that hydrocarbon generation has occurred in these rocks. PI ratios < 0.10 for 15 of 16 Simpson Group shale and limestone samples also indicate that significant generation of hydrocarbons has not occurred. The six Simpson Group samples with  $\leq$  2% TOC, and PI  $\geq$  0.17 may also indicate migrated hydrocarbons.

#### **Oil-source Rock Correlations**

Oil-source rock correlations are based on comparisons of oil geochemistry with the geochemistry of bitumens extracted from potential source rocks. Bitumens and oils are most easily correlated by comparison of saturated hydrocarbon and terpane (m/z=191) distributions, and carbon-isotope compositions of the saturated and aromatic hydrocarbon fractions. For oils (migrated bitumens) these parameters are affected by biodegradation in the reservoir and, possibly, by migration distance. Detailed discussions of the geologic/geochemical controls of bitumen and oil chemistries are beyond the scope of this paper. For an extensive introduction to these topics, refer to texts by Tissot and Welte (1978, 1984), and Hunt (1979); for indepth reviews of biomarkers, refer to Mackenzie (1984) and Peters and Moldowan (1993).

#### Oil Group 1

Comparison of the compositions of the Group 1 oils in the Forest City basin with the compositions of extracts of the Simpson Group shales in the basin show similar saturated hydrocarbon and terpane distributions (figs. 3A and 5A; 3D and 5D, respectively) and pristane/ $nC_{17}$  versus  $nC_{12}$ - $nC_{20}$  CPI (figs. 4A and 4B). These similarities suggest that the Simpson Group shales are the source rocks for these oils.

 $\delta^{13}C_{sat}$  of the Group 1 oils ranges from -26.8% to -31.3% of  $\delta^{13}C_{arom}$ , from -26.2% to -31.1% (table 2).  $\delta^{13}C$  compositions of the three extracts of the Simpson Group rocks from Wabaunsee County ( $\delta^{13}C_{sat}$  range from -25.6% to -26.4% of  $\delta^{13}C_{arom}$  from -25.4 to -26.7% to the similar to the positive end of this range. A better match in the observed ranges of  $\delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  for the Group 1 oils is with the range of composition observed ( $\delta^{13}C_{sat}$  from -24.5 to -31.1% of  $\delta^{13}C_{arom}$  from -23.7 to -31.3% in extracts of Simpson Group-equivalent rocks in Washington and Jackson counties, in east-central and

eastern Iowa (Hatch et al., 1991, their table 24-2). Because of the similar range of carbon isotope compositions of Forest City basin oils and rocks extract from Washington and Jackson counties, Iowa, we assume that our samples of the Simpson Group rocks in the Forest City basin show only a limited part of the range of organicmatter carbon-isotope compositions that are actually present.

The fact that these oils were generated from Middle Ordovician source rocks, but are now produced from stratigraphically higher units (e.g., Viola Limestone, Maquoketa Shale, and "Hunton Group" along the axis of the basin and from Pennsylvanian- and Mississippian-age horizons at the Easton and Vassar fields, respectively) indicates that vertical migration of these oils has occurred at many localities. Faults and fractures associated with development of the oil-bearing structures are the likely conduits for this migration.

The Group 1 oils are geochemically similar to Ordovician oils that are produced from other areas in the midcontinent region including the Iowa shelf (Hatch et al., 1991, their Group 1 oil), Salina-Sedgwick basin (Longman and Palmer, 1987; Hatch et al., 1987), and Anadarko basin (Burruss and Hatch, 1989, their Type 1 oil). Additionally, the Group 1 oils are geochemically similar to oils produced from the Ordovician Galena Group in the Illinois basin (Hatch et al., 1991, their Group 1 oils).

## Oil Group 2

Group 2 oils and extracts of the Chattanooga Shale have similar saturated hydrocarbon and terpane distributions (figs. 3B and 3E; 5B and 5E), pristane/ phytane (table 3), pristane/nC<sub>17</sub> versus nC<sub>12</sub>-nC<sub>20</sub> CPI (figs. 4A and 4B), and  $\delta^{\rm 13}C_{\text{sat}}$  and  $\delta^{\rm 13}C_{\text{arom}}$  measures, which indicate that the Chattanooga Shale is the most likely source for these oils. Although the Chattanooga Shale is thermally mature with respect to petroleum generation in the southern part of the Forest City basin, the ultimate origin of the Group 2 oils in the eastern production trend is subject to speculation. Some researchers (Rich, 1933; Price, 1980) favor a scenario in which these hydrocarbons were generated in the Anadarko and/or Arkoma basins in Oklahoma and underwent long-distance migration into Kansas.

Oils geochemically similar to the Group 2 oils are produced from: 1) the Cherokee Group, in the Cherokee basin to the south of the Forest City basin (Hatch et al., 1989); 2) Silurian, Devonian, and Mississippian rocks in the Anadarko basin in western Oklahoma and south-central and southwestern Kansas (Burruss and Hatch, 1989, their Type 2 oil); and 3) Silurian, Devonian, Mississippian, and Pennsylvanian rocks in the Illinois basin (Hatch et al., 1991, their Group 3 oils). Identified source rock for the

Group 2 equivalent oils on the Cherokee platform is the Chattanooga Shale (Hatch et al., 1989), in the Anadarko basin, the Woodford Shale (Burruss and Hatch, 1989), and in the Illinois basin, the New Albany Group (Hatch et al., 1991). The Chattanooga Shale, Woodford Shale, and the New Albany Group all are of Late Devonian-Early Mississippian age.

#### Oil Group 3

The Group 3 oil and extracts from the thin widespread Middle and Upper Pennsylvanian shales (offshore shales of Heckel, 1977) have similar saturated hydrocarbon and terpane distributions (figs. 3C and 3F; 5C and 5F, respectively) and similar  $\delta^{13}C_{sat}$  and  $\delta^{13}C_{arom}$  compositions which suggest that these shales are the most likely source rocks for this oil. The Group 3 oil is differentiated from the Group 2 oils by a lower pristane/phytane ratio (Group 3 oil, 0.9; Group 2 oil median, 1.6; table 3) and a higher pristane/nC<sub>17</sub> ratio (Group 3 oil, 3.1; Group 2 oil median, 0.56; table 3 and fig. 4A). This differentiation between Oil Groups 2 and 3 may become less clear, however, when additional Group 3 oils in the Forest City basin are identified, because extracts of Middle and Upper Pennsylvanian shales show significant ranges of pristane/ phytane (1.0 to 2.0, table 3) and pristane/ $nC_{17}$  (0.6–3.2, table 3, fig. 4B). It is unlikely, however, that Group 3 oils will be widespread in the Forest City basin, because the Pennsylvanian shales are thermally mature only in the southwestern corner of the basin.

The immature conditions generally found within the Pennsylvanian System in the Forest City basin indicate that Group 3 oils will be uncommon. However, local hot spots and a migration system where the source rock is located very near the potential reservoir may lead to commercial accumulations of Group 3 oils. Attention to formation evaluations of Pennsylvanian strata in existing oil fields may lead to overlooked pay zones containing this type of oil.

Oils geochemically similar to the Group 3 oils from the Forest City basin are produced from Pennsylvanianage reservoir rocks in the Anadarko basin in western Oklahoma and south-central and southwestern Kansas (Burruss and Hatch, 1989, their Type 3 oil). Burruss and Hatch (1989) identify the Middle and Upper Pennsylvanian marine black shales as the primary source rocks for the Type 3 oils in the Anadarko basin.

The saturated hydrocarbon and terpane distributions, and pristane/phytane, pristane/nC<sub>17</sub>, from the Marmaton Group coal are dissimilar to those of any of the 26 oils studied, which suggests that the Pennsylvanian coals in the Forest City basin are not effective source rocks for liquid hydrocarbons.

## Summary

- 1. Differences in saturated hydrocarbon and terpane (m/z =191) distributions, pristane/ $nC_{17}$ , and  $nC_{12}$ - $nC_{20}$  CPI of 26 Forest City basin oils demonstrate the existence of three distinct oil groups. Group 1 oils are produced primarily from lower Paleozoic reservoirs along the north-northeast-south-southwest-trending axis of the basin; Group 2 oils, primarily from Middle and Upper Pennsylvanian sandstones on the eastern flank of the basin; and the Group 3 oil, from the Kansas City Group in Wabaunsee County, Kansas.
- 2. Analyses of 123 representative rock samples from the Forest City basin demonstrate good to excellent potential to generate hydrocarbons for the Simpson Group shales and limestones, Chattanooga Shale, and Middle and Upper Pennsylvanian offshore marine shales.
- 3. In general, thermal maturity of organic matter in each stratigraphic interval increases from north to south. Organic matter in the Middle Ordovician strata is generally marginally mature to mature with respect to petroleum generation, whereas, organic matter in Pennsylvanian strata is generally immature to marginally mature.
- 4. Using the Jones (1987) classification, compositions of organic matter in shales from the Kansas City Group and the Chattanooga Shale are primarily organic facies BC, and, where thermally mature, these rocks should generate both oil and gas. Organic matter compositions in shales from the Cherokee, Marmaton. Pleasanton, Lansing, and Shawnee groups primarily represent organic facies C and CD, and, where thermally mature, these shales should primarily generate gas. Organic-matter compositions in the Simpson Group shales and limestones range from organic facies A to D, and, where thermally mature, the intervals containing organic matter with facies A, AB, and B compositions should generate oil.

- 5. Saturated hydrocarbon and terpane distributions for Group 1 oils are similar to those from extracts of Simpson Group rocks, suggesting that these rocks are the sources for the oils. Similar comparisons suggest that the source rocks for the Group 2 oils are in the Chattanooga Shale, whereas, source rocks for the Group 3 oil are most likely in Upper Pennsylvanian offshore marine shales.
- 6. The Forest City basin Group 1 oils are geochemically similar to other oil groups/types produced from the Anadarko basin, Salina basin, Iowa shelf, and the Illinois basin. The Forest City basin Group 2 oils are geochemically similar to oils produced from the Anadarko basin, Cherokee platform, and the Illinois basin, whereas, the Forest City basin Group 3 oil is similar to oils produced from the Anadarko basin.

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# Appendices

APPENDIX 1—Descriptive information for 26 oil samples from northwestern Missouri, southeastern Nebraska, and eastern Kansas. (Oil sample index map is fig. 2A. Oil groups are defined in text.)

Location map letter	Operator (if known) Field or well name (Producing horizon)	Location	Location map letter	Operator (if known) Field or well name (Producing horizon)	Location
	G 2 Oil		I	Group 1 Oils (con	<u>tinued)</u>
Kansas	Group 3 Oil	<u>s</u>	Nebraska		
A	Carter G. H. Davis A#15 Davis Ranch field (Kansas City Group)	SE SE SW sec. 33, T. 13 S., R. 10 E., Wabaunsee Co.	J	Nemaha # 1 Miles A Dawson field ("Hunton" and Viola Ls.)	SE NW NW sec. 11, T. 1 N., R. 14 E., Richardson Co.
	Group 2 Oi	ls	K	Reesman lease	SW sec. 17,
Kansas	GIOUP 2 G.	<u></u>		Falls City field ("Hunton")	T. 1 N., R. 16 E., Richardson Co.
В	Cities Service Yaege #12 Yaege field ("Hunton")	SE SE SE sec. 25, T. 11 S., R. 8 E., Riley Co.	К	Seiberson lease Falls City field ("Hunton")	SW sec. 17, T.1 N., R. 16 E., Richardson Co.
С	? Keegan lease Prairie Center A field (Cherokee)	SW NW SW sec. 34, T. 13 S., R. 22 E., Johnson Co.	K	Sand Rock lease Falls City field ("Hunton")	NW sec. 20, T. 1 N., R. 16 E., Richardson Co.
· D	T. I. E. Energy Kramer lease Gardner field (Bartlesville sand)	SW NW SE sec. 17, T. 14 S., R. 22 E., Johnson Co.	K	Harper #7 lease Falls City field ("Hunton")	E/2 N/2 NW sec. 20, T. 1 N., R. 16 E., Richardson Co.
E	Gene Ward Oil Co Andersen #2, Paola–Rantoul field	sec. 32, T. 16 S., R. 22 E., Miami Co.	K	Walldraff lease Falls City field ("Hunton")	SE SW sec. 20, T. 1 N., R. 16 E., Richardson Co.
F	(Squirrel sand) T. I. E. Energy	sec. 35,	K	Liniger lease Falls City field ("Hunton")	N/2 NE sec. 20, T. 1 N., R. 16 E., Richardson Co.
1	Polhamus #A-4 Goodrich-Parker field	T. 19 S., R. 22 E., Linn Co.	Kansas	( Hanton )	
G	(Bartlesville sand)  Cities Service Teeter #37 Teeter field (58204, Burbank sand)	NW sec.15, T. 23 S., R. 9 E., Greenwood Co.	L	Tomer Production Hustead lease Livingood field (Viola Limestone)	NE SW NW sec. 3, T. 1 S., R. 15 E., Brown Co.
Н	Amoco Headley "A" # 1- Winterscheid field (Bartlesville sand)	9 sec. 29/32, T. 23 S., R.14 E., Woodson Co.	M	Stuckey # 1 Aberle Sabetha field ("Hunton")	NW SE NW sec. 13, T. 2 S., R. 14 E., Nemaha Co.
	Group 10	<u>ils</u>	M	Carter # 1 Parli	NE NW SW sec. 27,
Missouri				Strahm field (Viola Limestone)	T. 2 S., R. 14 E., Nemaha Co.
I	Waters #1 Watkins Tarkio field (Bartlesville sand)	sec. 32, T. 65 N., R. 39 W., Atchison Co.	N	Petro-Lewis # 1 Potts McClain field (Viola Limestone)	C SE SE sec. 7, T. 4 S., R. 14 E., Nemaha Co. continued next pa

## Appendix 1 continued

Location map letter	Operator (if known) Field or well name (Producing horizon)	Location	Location map letter	Operator (if known) Field or well name (Producing horizon)	Location
	Group 1 Oils (cor	ntinued)	Í		
			P	Exxon #8 William O.	sec. 26,
N	Petro-Lewis # 1 Potts	C SE SE sec. 7,		Olsen	T. 15 S., R. 9 E.,
	McClain field	T. 4 S., R. 14 E.,		John Creek field	Morris Co.
	(Simpson Group)	Nemaha Co.		(Viola Limestone)	
О	Edmiston Oil	SE NW SW sec. 17,	Q	Ablah-Brandt #1 Greer	NW corn lot 3, sec. 19,
	Crook lease	T. 8 S., R. 21 E.,		Bosch field	T. 16 S., R. 8 E.,
	Easton field	Leavenworth Co.		(Hunton)	Morris Co.
	(McLouth sandstone)				
	, ,		R	Messman-Rinehart	NW NE NW sec. 23,
Α	Carter G. H. Davis A#11	SW NW NE sec. 33,		#1 Woodward	T. 16 S., R., 16 E.,
	Davis Ranch field	T. 13 S., R. 10 E.,		Vassar field	Osage Co.
	(Viola Limestone)	Wabaunsee Co.		(Mississippian limestone)	

APPENDIX 2—Descriptive information for 123 samples of Paleozoic-age rocks from southwestern Iowa, southeastern Nebraska, northwestern Missouri, and Eastern Kansas. (Rock-sample index map is fig. 2B.) The Shawnee Group is of Late Pennsylvanian (Virgilian) age; Lansing, Kansas City, and Pleasanton Groups, Late Pennsylvanian (Missourian) age; Marmaton and Cherokee Groups, Middle Pennsylvanian (Desmoinesian) age; Chattanooga Shale, Late Devonian–Early Mississippian age; Maquoketa Shale, Late Ordovician age; and Simpson Group, Middle Ordovician age.

Index map number	Operator Lease or well name	Description	Index map number	Operator Lease or well name	Description	
	Shawnee Group			Kansas City Group (continued)		
Iowa  1  Kansas	Iowa Geological Survey Bedford (IBC), SE sec. 4, T. 67 N., R. 34 W., Taylor Co.	Queen Hill Shale Member of Lecompton Limestone 3 core samples (IBCQHillA, B, C) 102.1-102.5 ft	5	Iowa Geological Survey Jefferson Quarry (IJCL) SE NW sec. 17, T. 77 N., R. 31 W., Adair Co.	Stark Shale Member of Dennis Limestone 1 representative core sample (IJCLStarkA) from 51.6-52.2 ft Hushpuckney Shale Member of Swope Limestone;1 representative	
2	Roxy Resources #1-14 Schneider NW SE SW sec.14, T. 1 S., R. 14 E., Nemaha Co.	1 cuttings sample (970-1,000 ft)	6	Iowa Geological Survey Riverton (IRC) SE SE SW sec. 20,	core sample (IJCLHushA) from 85.3-85.8 ft Hushpuckney Shale Mem- ber of Swope Limestone 4 core samples; (IRCHP	
3	Heller #1 Stice NW sec. 11, T. 14 S., R. 8 E., Wabaunsee Co.	Heebner Shale Member of Oread Limestone 1 cuttings sample (1,180-1,190 ft)	1	T. 67 N., R. 41 W., Fremont Co. Iowa Geological Survey Bedford (IBC) SE sec. 4,	1-4); 1,036.6-1,039.2 ft  Hushpuckney Shale Member of Swope Limestone 4 core samples (IBCHP	
Iowa	<u>Lansing G</u>	roup		T.67 N., R. 34 W., Taylor Co.	1-4) 546.3-548.5 ft	
4	Iowa Geological Survey Hillsdale/Buffington (IHC), sec. 22, T. 72 N., R., 42 W., Mills Co.	Eudora Shale Member of Stanton Limestone 1 core sample (IHC Eudora) 100.2-100.5 ft	Missour 7	R. M. Coveney 12 NE sec. 5, T. 49 N., R. 33 W., Jackson Co.	Muncie Creek Shale Member of Iola Limestone 3 core sample (RC12) (about 30 ft below surface)	
Kansas 2	Roxy Resources #1-14 Schneider NW SE SW sec. 14, T. 1 S., R. 14 E.,	1 cuttings sample (1,330-1,360 ft)	8	WM-10, SE SE sec. 4, T. 59 N., R. 34 W., Andrew Co.	Wea Shale Member of Cherryvale Shale; 2 core samples; (WM10Wea 1 an 2) 415.5-421.8 ft	
3	Nemaha Co. Heller #1 Stice NW sec. 11,	1 cuttings sample (1,610-1,630 ft)	7	R. M. Coveney 15 NE sec. 5, T. 49 N., R. 33 W., Jackson Co.	Stark Shale Member of Dennis Limestone; 4 core samples (RC15) about 12 t below surface	
	T. 14 S., R. 8 E., Wabaunsee Co.	Group	9	WM-9, NE SE sec. 9, T. 53 N., R. 36 W.,	Hushpuckney Shale Member of Swope Limestone; I representative core sample	
Iowa				Platte Co.	(WM9HP4) from 302.50- 304.3 ft	
4	Iowa Geological Survey Hillsdale/Buffington (IHC sec. 22, T. 72 N., R. 42 W., Mills Co.	Quindaro Shale Member of C)Wyandotte Limestone 3 core samples (IHC QuinA, B, C) 166.1-166.4 ft	7	R. M. Coveney 5 center, sec. 18, T. 49 N., R. 33 W., Jackson Co.	Hushpuckney Shale Member of Swope Limestone 1.8 channel sample (RC5) from limestone mine abou 100 ft below surface	

Index map number	Operator Lease or well name	Description	Index map number	Operator Lease or well name	Description
	Kansas City Group	(continued)			
Kansas			Missouri	i	
2	Roxy Resources #1-14 Schneider NW SE SW sec. 14, T. 1 S., R. 14 E., Nemaha Co.	1 cuttings sample (1,440-1,470 ft)	15	USAF MC 121 SW sec. 1, T. 40 N., R. 32 W., Bates Co.	Anna Shale Member of Pawnee Limestone, 2 core samples (MC121-2, MC121-3) 62.3-65.1 ft Lexington Coal Member,
10	Mid Gulf #1 Caudle Farm NE NE NW sec. 28, T. 4 S., R. 20 E., Doniphan Co.	4 cuttings samples (1,200-1,220, 1,365-1,375, 1,545-1,555 and 1,800-1,810 ft)	Kansas	D D	1 core sample (MC121-4; 73.3-73.8 ft)
11	Kansas Geological Survey Edmonds #1A SE NW SW sec. 35, T. 9 S., R. 22 E.,	Dennis Limestone 4 representative core sam- ples (EDS1A 3-16) from	2	Roxy Resources #1-14 Schneider NW SE SW sec. 14, T. 1 S., R. 14 E., Nemaha Co.	1 cuttings sample (1,540-1,570 ft)
12	Leavenworth Co.  Carter G. H. Davis #2-A SW NE SW sec. 33, T. 13 S., R. 10 E.,	401.35 to 403.37 ft  Stark Shale Member of Dennis Limestone 10 representative core	3	Heller #1 Stice NW sec. 11, T. 14 S., R. 8 E., Wabaunsee Co.	1 cuttings sample (1,780-1,800 ft)
	Wabaunsee Co.	samples (S2-S13) from 1,862.7 to 1,864.4 ft		Marathon Oil Co. J. W. Martindell #50,	Excello shale member of Fort Scott Limestone
3	Heller #1 Stice NW sec. 11, T.14 S., R. 8 E.,	Stark Shale Member of Dennis Limestone 1 cuttings sample (1,660-		SW sec. 31, T. 23 S., R. 10 E., Greenwood Co.	1 core sample (349-3; 2,095.1-2,096.1 ft)
	Wabaunsee Co.	1,680 ft) Hushpuckney Shale Mem-		Cherokee C	<u>Froup</u>
13	Kansas Geological Survey Woodward #5A	ber of Swope Limestone 1 cuttings sample (1,680- 1,700 ft) Stark Shale Member of Dennis Limestone		Roxy Resources #1-14 Schneider NW SE SW sec. 14, T. 1 S., R. 14 E.,	5 cuttings samples (1,690-1,730, 1,900-1,940, 1,980-2,000, 2,230-2,240 and 2,330-2,340 ft)
	NE SE SW SE sec. 12, T. 25 S., R. 21 E., Bourbon Co.	6 representative core samples (W5A3-W5A16) from 49.64-54.03 ft	10	Nemaha Co. Mid Gulf #1 Caudle Farm NE NE NW sec. 28,	2 cuttings samples (2,020-2,030 ft, 2,110-2,120 ft)
	<u>Pleasanton (</u>	<u>Group</u>		T. 4 S., R. 20 E., Doniphan Co.	·
Kansas					4
	Heller #1 Stice NW sec. 11, T. 14 S., R. 8 E., Wabaunsee Co.	2 cuttings samples (1,710-1,730 ft, 1,760-1,770 ft)		Heller #1 Stice NW sec. 11, T. 14 S., R. 8 E., Wabaunsee Co.	1cuttings sample (1,850-1,870 ft)
	Marmaton G	Group		Marathon Oil Co. J. W. Martindell #50	Unnamed coal member of Cabaniss Formation;
	wa		SW sec. 31, T. 23 S., R. 10 E., Greenwood Co.	1 core sample (349-38; 2,213.0-2,213.1 ft) Unnamed shale members of Cabaniss Formation; 3 core samples (349-46, 2,225.0-2,226.4 ft; 349-57, 2,252.8	
	Clarke Co.	Little Osage Shale Member of Fort Scott Limestone 1 core sample (CP37-24, 205.3-206.4 ft)			2,253.9 ft; 349-69, 2,288.7 2,289.3 ft)  continued next pa

Index map	Operator Lease or well name	Description	Index map number	Operator Lease or well name	. Description
number			number		

## **Chattanooga Shale**

<u>Onutumoogu Shure</u>				
Nebrask	a			
17	Big J Production #1 Harper-Sibbernsen SE NE NW sec. 20, T. 1 N., R. 16 E., Richardson Co.	1 core sample (HS2196) 2,196 ft		
Kansas				
2	Roxy Resources #1-14 Schneider NW SE SW sec. 14, T. 1 S., R. 14 E., Nemaha Co.	2 cuttings samples (2,540-2,550, 2,680-2,690 ft)		
18	Pendleton #1 Hermesch NW SE SE sec. 9, T. 5 S., R. 13 E., Nemaha Co.	1 cuttings sample (2,730-2,740 ft)		
12	Carter G. H. Davis #2-A SW NE SW sec. 33, T. 13 S., R. 10 E., Wabaunsee Co.	2 core samples (DR2977, DR2980) from basal 3.0 ft of shale, 2,977-2,980 ft		
3	Heller #1 Stice NW sec. 11, T. 14 S., R. 8 E., Wabaunsee Co.	1 cuttings sample, 2,010- 2,140 ft		
19	Shell #1 Farr NW SE SW sec. 7, T. 20 S., R. 12 E., Lyon Co.	1 core sample (F2427) 2,427 ft		
19	Shell #1-9 Friday NE SW SW sec. 9, T. 20 S., R. 12 E., Lyon Co.	1 core sample (F2566), 2,566 ft		
20	ERDA Stauffer #1 NE sec. 20, T. 23 S., R. 12 E., Greenwood Co.	5 core samples (STA2313, STA2315, STA2316, STA2318, STA2319), from basal 9.0 ft of shale, 2,312- 2,322 ft		
21	ERDA Bock #1 SW NE NE sec. 15, T. 23 S., R. 12 E., Greenwood Co.	6 core samples (BOC2164, BOC2166, BOC2168, BOC2170, BOC2172, BOC2174) from basal 12.2 ft of shale, 2,164-2,174.2 ft		

## Magnakata Shala

	<u>Maquoketa S</u>	<u>Shale</u>
Kansas		
2	Roxy Resources #1-14 Schneider NW SE SW sec. 14, T. 1 S., R. 14 E., Nemaha Co.	1 cuttings sample (3,410-3,430 ft)
12	Carter G. H. Davis #2-A SW NE SW sec. 33, T. 13 S., R. 10 E., Wabaunsee Co.	1 core sample (DR3200 ) representing basal 13 ft of shale (3,195-3,208 ft)
3	Heller #1 Stice NW sec. 11, T.14 S., R. 8 E., Wabaunsee Co.	1 cuttings sample (2,162-2,165 ft)
	Simpson G	<u>roup</u>
Missouri	i	
22	#1 McCartney SE SE SE sec. 5, T. 64 N., R. 39 W., Atchison Co.	5 cuttings and sidewall core samples (clay shales at 3,126, 3,130-3,140, 3,142, 3,156, and 3,173 ft)
Kansas		
2	Roxy Resources #1-14 Schneider NW SE SW sec. 14, T. 1 S., R. 14 E., Nemaha Co.	1 cuttings sample (3,700-3,720 ft)
10	Mid Gulf #1 Caudle Farm NE NE NW sec. 28, T. 4 S., R. 20 E., Doniphan Co.	1 cuttings sample (2,725-2,730 ft)
12	Carter G. H. Davis #2-A SW NE SW sec. 33, T. 13 S., R. 10 E., Wabaunsee Co.	15 core samples (DR3298.5, DR3307.5, DR 3311.5, DR3312.5, DR3313.5, DR3314.8, DR3315.5, DR3320.4, DR3321, DR3322.5, DR3323.5, DR3326.2, DR3358.2, DR3358.7, DR3381.4; 3,298.5-3,387 ft)

APPENDIX 3—Organic-carbon contents and Rock-Eval® analysis for 87 samples from cores of Paleozoic-age rocks from southwestern Iowa, southeastern Nebraska, northwestern Missouri, and northeastern Kansas. [Sample descriptions are listed in appendix 2. (Rock-sample index map is fig. 2B.) The Shawnee Group is of Late Pennsylvanian (Virgilian) age; Lansing and Kansas City Groups, Late Pennsylvanian (Missourian) age; Marmaton and Cherokee Groups, Middle Pennsylvanian (Desmoinesian) age; Chattanooga Shale, Late Devonian–Early Mississippian age; Maquoketa Shale, Late Ordovician age; and Simpson Group, Middle Ordovician age.]

Location map number	Sample number	Organic carbon (weight %)	T <sub>max</sub> (°C)	Hydrogen <sup>b</sup> index (mg/g)	Oxygen <sup>c</sup> index (mg/g)	Genetic <sup>d</sup> potential (mg/g)	Production <sup>e</sup> index
Town				Shawnee Grou	пБ		
Iowa							
1	IBCQHillA	10.1	411	180	63	19	0.04
	<b>IBCQHillB</b>	21.0	415	235	52	51	0.04
	IBCQHillC	5.2	420	220	50	12	0.03
<b>T</b>				Lansing Grou	ıр		
Iowa							
4	IHCEudora	15.2	423	130	52	21	0.03
Iowa			<u>k</u>	Kansas City Gr	oup		
4	IHCQuinA	5.7	418	145	55	8.4	0.03
	IHCQuinB	16.8	410	270	61	47	0.03
	IHCQuinC	7.5	414	225	58	17	0.04
5	IJCLStarkA	18.4	412	360	43	67	0.02
	<b>IJCLHushA</b>	27.3	414	330	34	91	0.02
6	IRCHP1	7.4	425	285	54	29	0.01
	IRCHP2	9.5	422	295	58	29	0.01
	IRCHP3	30.8	420	305	47	97	0.02
	IRCHP4	15.8	419	360	46	58	0.02
1	IBCHP1	4.5	428	270	48	13	0.02
	IBCHP2	24.3	426	195	63	49	0.02
	IBCHP3	18.5	429	185	65	35	0.03
	IBCHP4	5.6	433	150	48	8.6	0.02
Missouri							
7	RC12	13.2	417	360	26	48	0.02
8	WM10Wea1	1.8	424	22	21	0.5	0.10
	WM10Wea2	1.5	420	13	20	0.3	0.10
7	RC15	24.4	420	295	22	75	0.03
9	WM9HP	13.3	429	230	50	32	0.03
7	RC5	15.4	424	440	24	71	0.05
Kansas							
11	EDS1A16	4.8	432	400	18	19	0.02
	EDS1A12	29.3	422	505	26	150	0.02
	EDS1A5	25.9	422	425	25	110	0.02
	EDS1A3	11.0	425	320	36	40	0.02
12	S2-1	29.5	423	430	9	140	0.04
-	S3-1	2.1	431	340	13	8.8	0.08
	S4-1	21.4	423	435	13	100	0.18
	S5	14.4	425	465	16	72	0.08
	\$7	2.2	429	170	37	4.2	0.07
	S8T	2.2 17.9	425	390	15	4.2 76	0.09

Location map number	Sample number	Organic carbon (weight %)	T <sub>max</sub> (°C)	Hydrogen <sup>b</sup> index (mg/g)	Oxygen <sup>c</sup> index (mg/g)	Genetic <sup>d</sup> potential (mg/g)	Production <sup>e</sup> index
			Kansas Ci	ty Group (cont	inued)		
	S9B	0.9	425	86	85	0.8	0.15
	S11	13.5	427	360	16	53	0.07
	S12B	23.6	424	340	22	89	0.10
			430	210	48	3.7	0.08
	S13	1.6			98	1.6	0.03
13	W5A3	1.1	438	140			0.05
	W5A4	4.9	439	210	65	10	
	W5A6	16.9	436	360	36	63	0.05
	W5A10	23.2	431	370	39	89	0.03
	W5A12	13.5	429	230	55	33	0.05
	W5A16	7.6	438	200	52	16	0.05
Iowa				Marmaton Gr	<u>oup</u>		
		40.0	422	275	<b>51</b>	52	0.02
14	CP3719A	18.9	422	275	51	53	0.02
	CP3719B	3.5	433	68	75	2.5	0.04
	CP3719C	15.2	424	230	72	36	0.02
	CP3724	7.3	436	34	44	2.7	0.06
Missouri							
15	MC121-2	19.7	439	400	7	83	0.05
	MC121-3	9.9	441	220	7	23	0.07
	MC121-4	61.0	436	230	5	150	0.04
Kansas							
16	349-3	21.6	440	280	6	64	0.06
**				Cherokee Gr	<u>oup</u>		
Kansas							
16	349-38	47.0	445	230	5	120	0.05
	349-46	14.9	449	260	11	41	0.05
	349-57	8.1	446	170	14	15	0.09
	349-69	9.0	447	170	21	18	0.09
				Chattanooga S	<u>Shale</u>		
Nebraska	1152106	2.9	438	490	22	3.8	*
17	HS2196	2.8	430	490	22	5.6	
Kansas							
12	DR2977	1.9	434	240	25	5.6	0.23
	DR2980	3.4	438	390	8	15	0.10
19	F2427	4.4	442	390	13	17	*
19	F2566	2.6	440	210	15	5.4	*
	STA2313	4.1	443	320	29	15	0.11
20		4.1	445	370	18	17	0.11
20	STA2315						
20	STA2315 STA2316		442	290	57	10	0.12
20	STA2316	3.1	442 444	290 390	57 14		
20	STA2316 STA2318	3.1 .5.0	444	390	14	22	0.09
20	STA2316	3.1					

continued next page

Location map number	Sample number	Organic carbon (weight %)	T <sub>max</sub> (°C)	Hydrogen <sup>b</sup> index (mg/g)	Oxygen <sup>c</sup> index (mg/g)	Genetic <sup>d</sup> potential (mg/g)	Production <sup>e</sup> index
			Chatta	anooga Shale (c	continued)		
	BOC2168	4.0	444	350	26	16	0.10
	BOC2170	3.2	443	350	30	12	0.11
	BOC2172	4.2	444	320	46	15	0.09
	BOC2174	3.6	445	290	53	12	0.10
			<u>N</u>	Iaquoketa Shal	<u>le</u>		
Kansas							
12	DR3200	0.3	425	90	50	0.2	0.01
			<u> </u>	Simpson Group	1		
Kansas				_			
12	DR3298.5	0.6	432	220	24	1.6	0.17
	DR3307.5	0.3	443	330	42	1.4	0.22
	DR3311.5	0.5	437	380	40	1.5	0.02
	DR3312.5	1.0	447	430	16	4.8	0.06
	DR3313.5	15.8	444	935	4	150	0.02
	DR3314.8	1.0	443	360	27	3.9	0.05
	DR3315.5	1.4	440	260	23	4	0.07
	DR3320.4	0.8	438	150	19	1.3	0.11
	DR3321.0	0.9	437	370	28	3.7	0.06
	DR3322.5	0.7	433	180	32	1.6	0.24
	DR3323.5	0.5	435	235	70	1.7	0.27
	DR3326.2	0.8	436	385	38	3.6	0.20
	DR3358.2	11.2	442	515	3	60	0.03
	DR3358.7	6.2	439	425	5	27	0.03
	DR3381.4	1.8	439	11	14	0.3	0.34

<sup>\*</sup> indicates no data

HC = hydrocarbons; TOC = total organic carbon

Analyses for samples from the Stark Shale Member of the Dennis Limestone (S2-1 through S13) were previously listed in Hatch and Leventhal (1992, their table 3); for Marmaton Group samples (MC121-2 to MC121-4 and 349-3) and for Cherokee Group samples (349-38 though 349-69), Hatch et al. (1984, their appendix 3, and 1989, their appendix 3); and for Chattanooga Shale samples (DR2977, DR2980, STA2313 through STA2319, and BOC2164 through BOC2174), Hatch et al. (1989, their appendix 3).

a. Temperature at which the yield of pyrolysis products (S<sub>2</sub>) is at a maximum

b. S<sub>2</sub>/TOC, mgHC/gTOC

c. S<sub>3</sub>/TOC, mgCO<sub>2</sub>/gTOC

d.  $S_1 + S_2$ , mgHC/g sample

e.  $S_1/(S_1 + S_2)$ 

APPENDIX 4—Organic-carbon contents, Rock-Eval® analysis, and vitrinite-reflectance (R<sub>o</sub>) measurements for 36 samples from cuttings and sidewall cores of Paleozoic-age rocks from northwestern Missouri and northeastern Kansas. [Simpson Group samples 3126, 3142, 3156, and 3173 are from sidewall cores. All other samples are from cuttings. Sample descriptive information is listed in appendix 2. (Rock-sample index map is fig. 2B.) The Shawnee Group is of Late Pennsylvanian (Virgilian) age; Lansing, Kansas City, and Pleasanton Groups, Late Pennsylvanian (Missourian) age; Marmaton and Cherokee Groups, Middle Pennsylvanian (Desmoinesian) age; Chattanooga Shale, Late Devonian–Early Mississippian age; Maquoketa Shale, Late Ordovician age; and Simpson Group, Middle Ordovician age.]

Location map number	Sampled interval	Organic carbon (weight %)	T <sub>max</sub> (°C)	Hydrogen <sup>b</sup> index (mg/g)	Oxygen <sup>c</sup> index (mg/g)	Genetic <sup>d</sup> potential (mg/g)	Production <sup>e</sup> index	Vitrinite reflectance (%R <sub>0</sub> )
Vancas				Shawnee	Group			
Kansas								
2	970-1000	7.2	416	365	22	27	0.04	0.39?
3	1180-1190	10.4	435	200	*	22	0.04	0.55
Kansas				<b>Lansing</b>	<u>Group</u>			
2	1330-1360	10.0	417	415	20	43	0.04	0.40?
3	1610-1630	2.0	444	19	*	0.43	0.14	0.57
Kansas				Kansas Cit	y Group			
2	1440-1470	9.7	419	440	18	45	0.04	*
10	1200-1220	9.0	426	200	8	18	0.04	0.51
	1365-1375	1.0	432	40	53	0.51	0.24	0.59
	1545-1555	3.5	433	110	11	3.9	0.06	0.51
	1800-1810	3.5	433	85	9	3.2	0.07	0.57
3	1660-1680	14.0	434	220	*	32	0.04	*
	1680-1700	25.0	437	295	*	76	0.04	0.55
Kansas				<u>Pleasanton</u>	Group			
3	1710-1730	8.0	440	140	*	12	0.06	*
	1760-1770	4.4	444	120	*	5.7	0.06	0.60
Kansas				Marmaton	Group			
2	1540-1570	2.6	429	38	18	1.2	0.20	*
3	1780-1800	1.1	444	60	*	0.8	0.21	*
Kansas				Cherokee	Group			
2	1690-1730	11.5	421	410	12	49	0.04	0.47
_	1900-1940	5.9	428	180	11	11	0.05	*
	1980-2000	2.9	430	70	13	2.2	0.07	*
	2230-2240	4.9	430	120	20	6.1	0.05	*
_	2330-2340	3.2	431	93	18	3.2	0.06	0.54
0	2020-2030	2.3	434	64	22	1.6	0.10	0.56
2	2110-2120	1.1	434	300	23	3.4	0.01	*
3	1850-1870	1.8	460	26	*	0.6	0.16	0.70
Kansas				Chattanoog	ga Shale			
2.	2540-2550	1.9	433	35	19	0.7	0.15	*
	2680-2690	1.9	431	41	19	0.9	0.10	0.58
18	2730-2740	1.4	433	67	42	1.2	0.15	0.63
3	2010-2140	2.1	444	75	*	1.7	0.08	0.60

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#### Appendix 4 continued

Location map number	Sampled interval	Organic carbon (weight %)	T <sub>max</sub> (°C)	Hydrogen <sup>b</sup> index (mg/g)	Oxygen <sup>c</sup> index (mg/g)	Genetic <sup>d</sup> potential (mg/g)	Production <sup>e</sup> index	Vitrinite reflectance (%R <sub>0</sub> )
Kansas				Maquoket	ta Shale			
2	3410-3430	3.1	430	70	17	2.5	0.11	*
3	2162-2165	1.5	447	60	*	1.0	0.11	*
Missouri				Simpson	Group			
22	3126	1.5	442	690	70	11	0.08	*
	3130-3140	14.5	444	1150	7	180	0.05	*
	3142	2.0	436	620	40	13	0.08	*
	3156	1.1	439	370	120	4.5	0.08	*
	3173	0.5	437	350	240	1.9	0.07	*
Kansas								
2	3700-3720	8.1	441	810	11	68	0.03	*
10	2725-2730	1.1	446	490	27	5.8	0.05	*

<sup>\*</sup> indicates no data

HC = hydrocarbons; TOC = total organic carbon

a. Temperature at which the yield of pyrolysis products (S<sub>2</sub>) is at a maximum b. S<sub>2</sub>/TOC, mgHC/gTOC c. S<sub>3</sub>/TOC, mgCO<sub>2</sub>/gTOC d. S<sub>1</sub> + S<sub>2</sub>, mgHC/g sample e. S<sub>1</sub>/(S<sub>1</sub> + S<sub>2</sub>)



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